



Nano Miniaturization of a Hydrogen Plant

Evan Jones, Jamie Holladay, James Cao,
Gordon Xia, Rob Dagle
Pacific Northwest National Laboratory
DARPA MTO

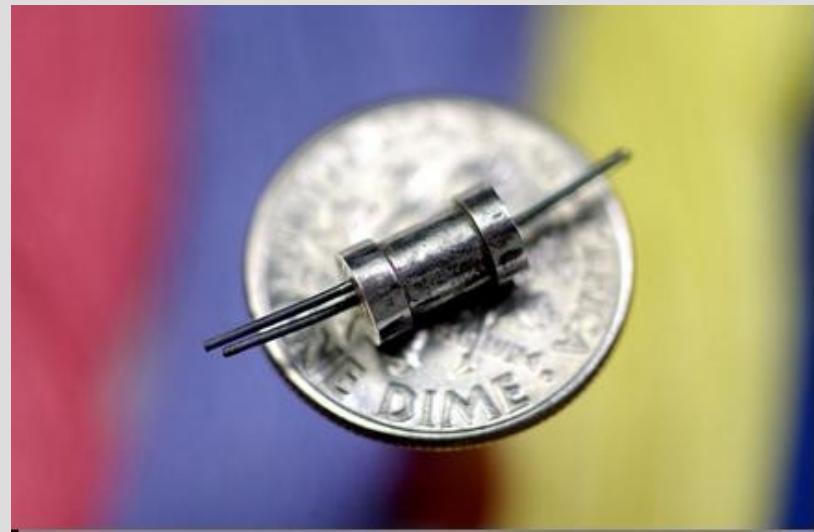


Introduction – Brief Overview

- ▶ Objective: Hydrogen production for micro power supply
- ▶ Background
 - Energy density hydrocarbons >> secondary batteries
 - Methanol = $5.6 \text{ kW}_t\text{-hr/kg}$, lithium-ion batteries = $0.12 \text{ kW}_e\text{-hr/kg}$
 - 10% efficient conversion = 4.6 times energy density increase
- ▶ Result: Micro-scale fuel processor
 - **One Billionth (10^{-9})** volume and process capacity
 - Reproduces major unit operations of a hydrogen plant
 - Fits on a dime

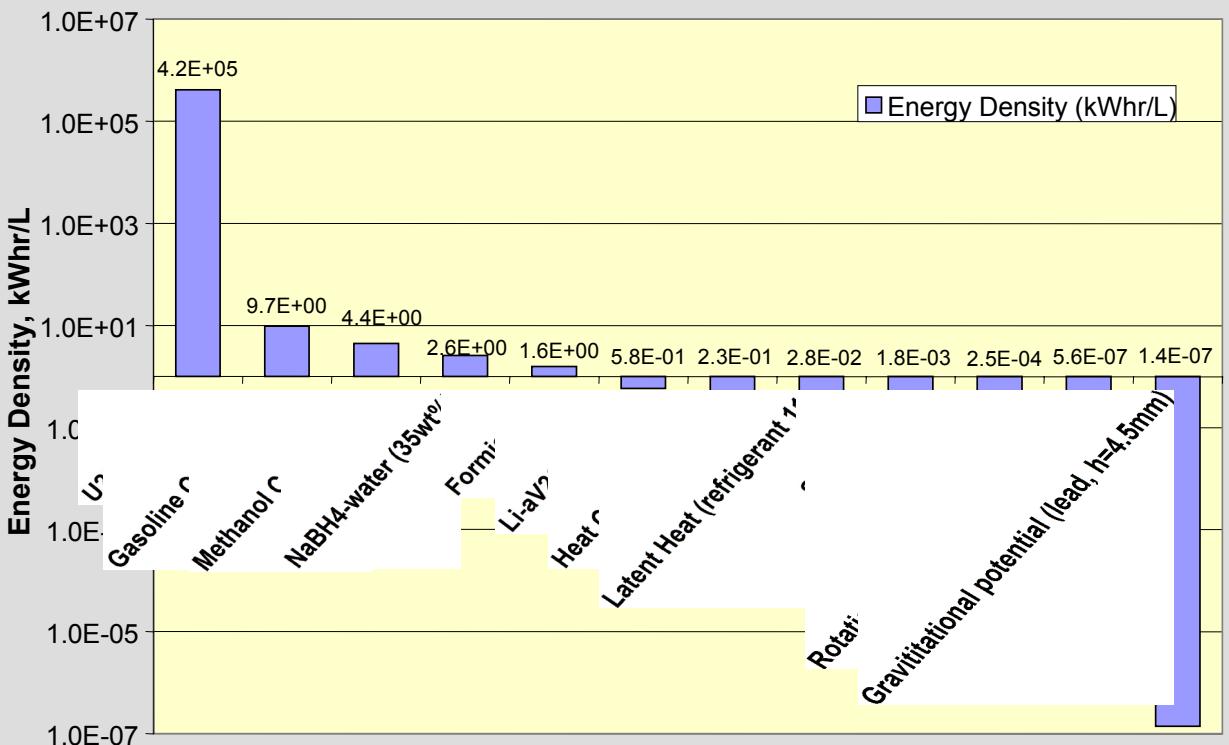
Microscale Reactor

- ▶ Hydrogen plant on a dime—the microreactor vision!
- ▶ New basis for engineering systems
 - Hydrogen distribution
 - Power supplies
 - Miniature products
 - Chemical labs
- ▶ Lessons learned
 - Keep it simple
 - Be flexible
 - Build and test
 - Metal can work



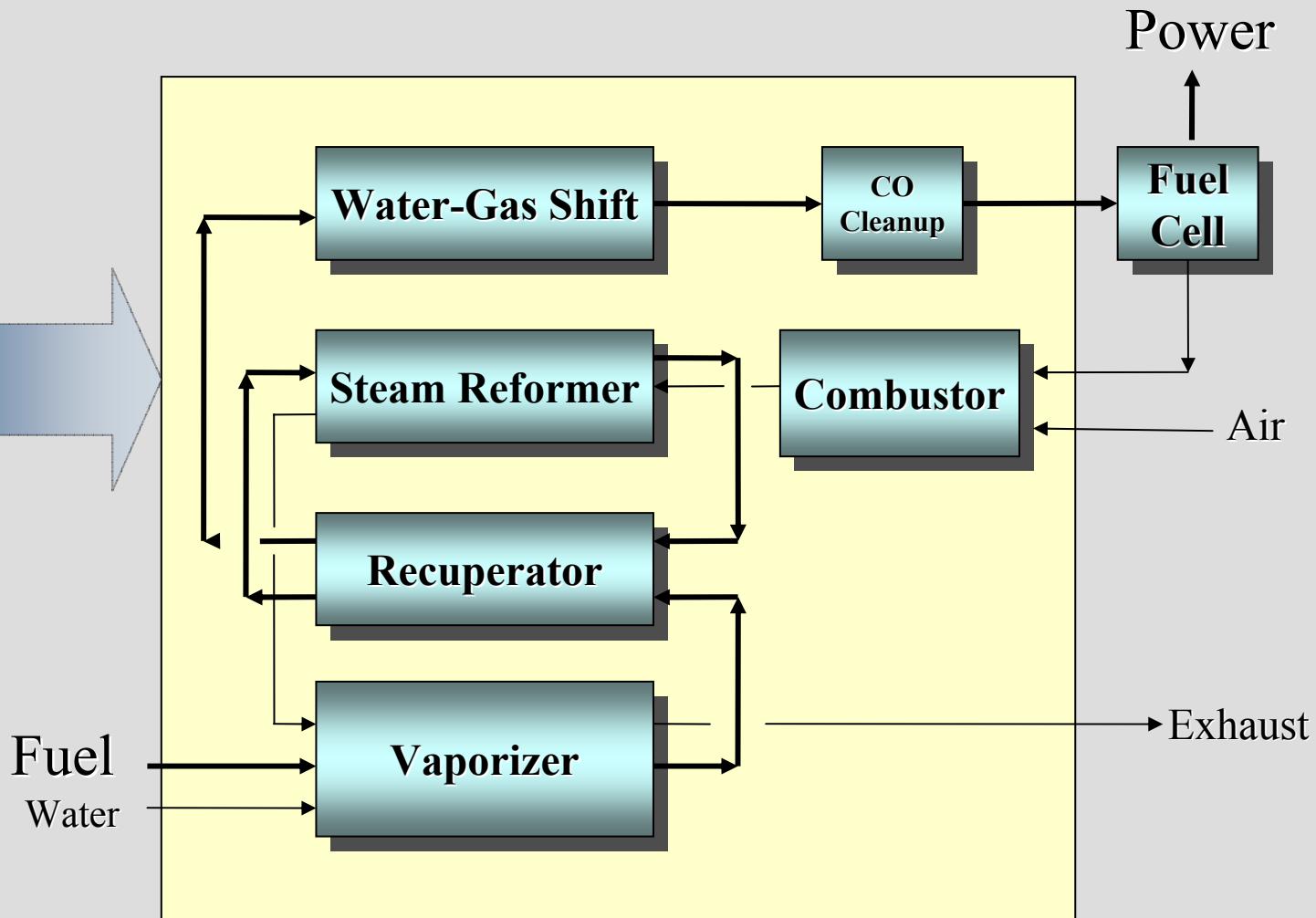
Energy Density of Some Potential Storage Devices

- Except for the battery, these storage devices will need a conversion device to make electrical power

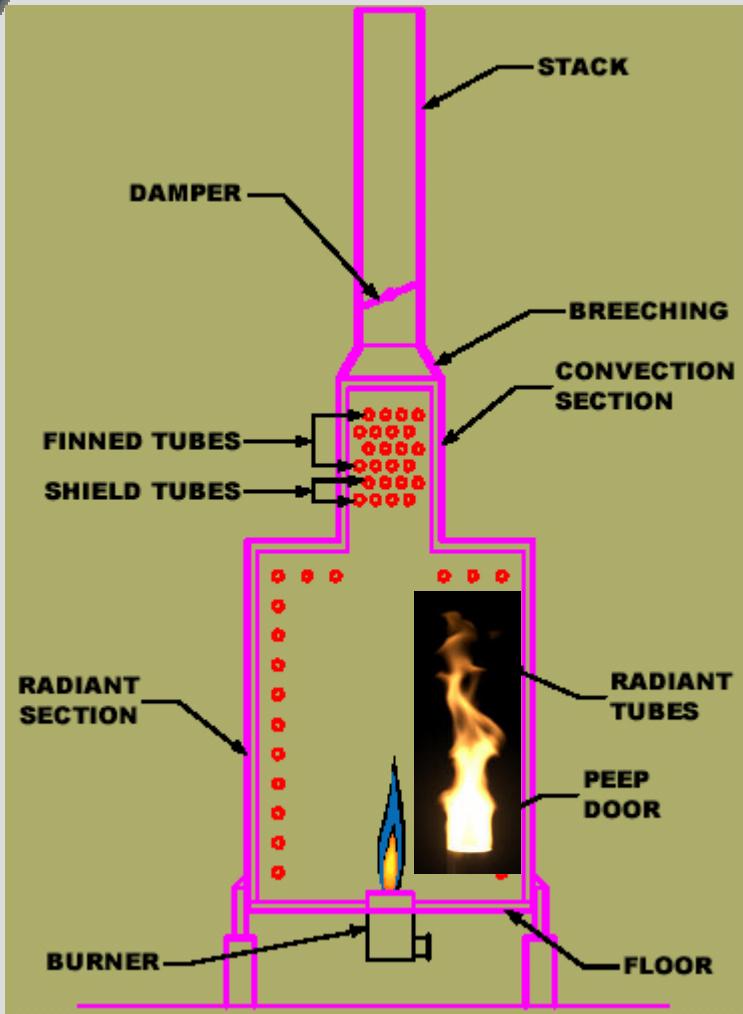


Adapted from Paul B. Koeneman et al. J. Microelectromechanical Systems vol. 6 no.4 1997 p 355

Hydrogen Production



Conventional Technology: Tube Box Heaters



How Do You Miniaturize Conventional Technology?

- ▶ Combine:
 - Chemical processing +
 - Microelectronic fabrication +
 - Nano technology

Scale-Down Approach

- ▶ Identify significant unit operations
- ▶ Identify controlling mechanisms/opportunities
 - How will scale-down affect them?
 - Does nano technology enable a new approach?
 - Reactions
 - Materials
 - Designs
- ▶ Design, build, test
 - Keep it simple – both the design and fabrication
 - Keep the end in mind
 - How will it be tested?
 - Is mass fabrication feasible?
 - Try multiple designs
- ▶ Model
- ▶ Repeat: design, build, test, model

Miniaturization Issues

Unit Op	Opportunity	Enabling Tech	Comments
Reforming reactor	<ul style="list-style-type: none">- heat transfer- mass transfer- slow kinetics	<ul style="list-style-type: none">- microchannel- improved catalysts	<ul style="list-style-type: none">- Increase heat and mass transfer- Fast kinetics
Heat source	<ul style="list-style-type: none">- quenching- T control	<ul style="list-style-type: none">- catalytic combustion- microchannel	<ul style="list-style-type: none">- Quenching is not an issue with- Catalytic combustion and microchannels improve T control
PSA	<ul style="list-style-type: none">- CO removal	<ul style="list-style-type: none">- methanation	<ul style="list-style-type: none">- Thermal runaway mitigated with microchannel reactors- Passive CO removal
Heat exchanger	<ul style="list-style-type: none">- thermal loss- heat transfer	<ul style="list-style-type: none">- design, fuel, and material selection	<ul style="list-style-type: none">- Tight unit op integration- Low-temperature reactions- Appropriate materials

Miniaturization Focus: High Efficiency/Minimize Heat Loss

- ▶ Low-temperature reactors
 - Fuel Selection
 - Catalyst
 - Reactor design
- ▶ Low volume and surface area
- ▶ Thermal integration
 - Design: put the heat only where it is needed
- ▶ Insulation
 - Material selection
 - Vacuum package?

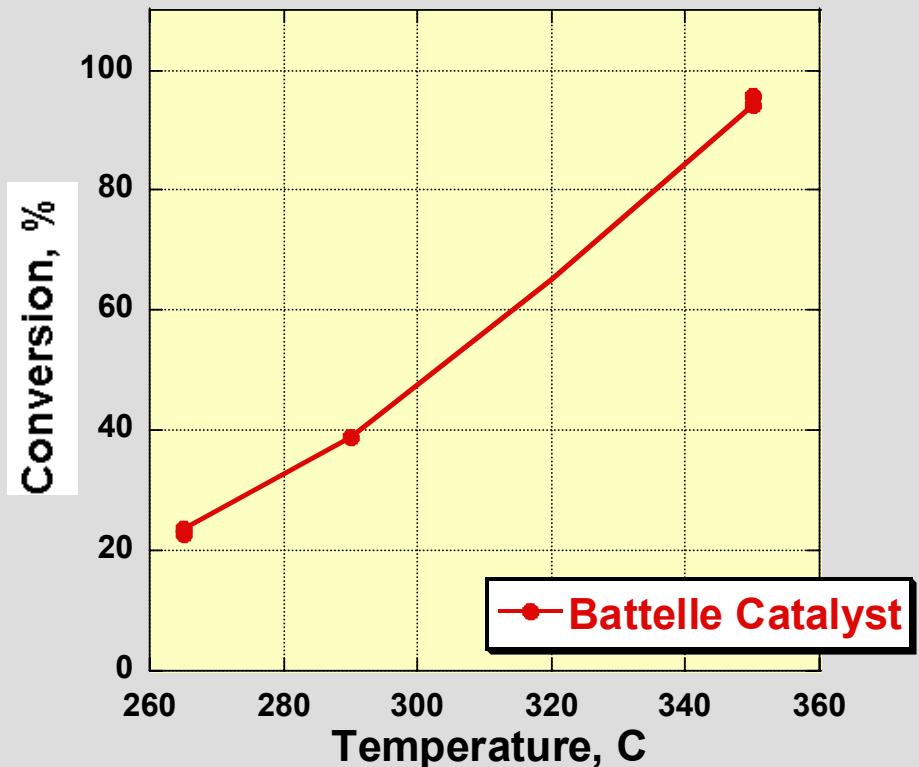
Fuel Selection

- Methanol mix has a low reforming T and high energy density

Fuel	Reaction T (°C)	Specific Energy (pure)	Specific Energy (mix)
Methanol	250-350	5.6 kW _t -hr/kg	1:1 S:C ~3.2 kW _t -hr/kg
C8-C12 hydrocarbon	650-800	12.5 kW _t -hr/kg	2:1 S:C ~3 kW _t -hr/kg
Methanol	DMFC	5.6 kW _t -hr/kg	3M 0.55 kW _t -hr/kg
Formic Acid	Direct FA FC	1.6 kW _t -hr/kg	10M 0.55 kW _t -hr/kg

Methanol-Specific Battelle Catalyst

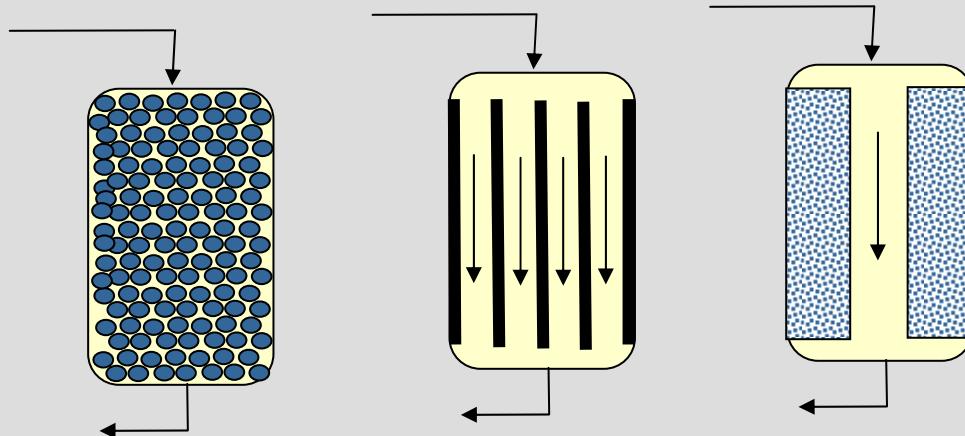
- ▶ Reactor Conditions
 - Atmospheric pressure
 - Temperature 260-350°C
- ▶ **24,000-50,000 hr⁻¹ GHSV**
- ▶ Catalyst Life >1000 hr
- ▶ Non-pyrophoric
- ▶ Typical Composition
(dry gas)
 - H₂ = 72-74%
 - CO₂ = 24-26%
 - CO = <1%
 - Equilibrium CO at 350°C = 5.4%



Design Options

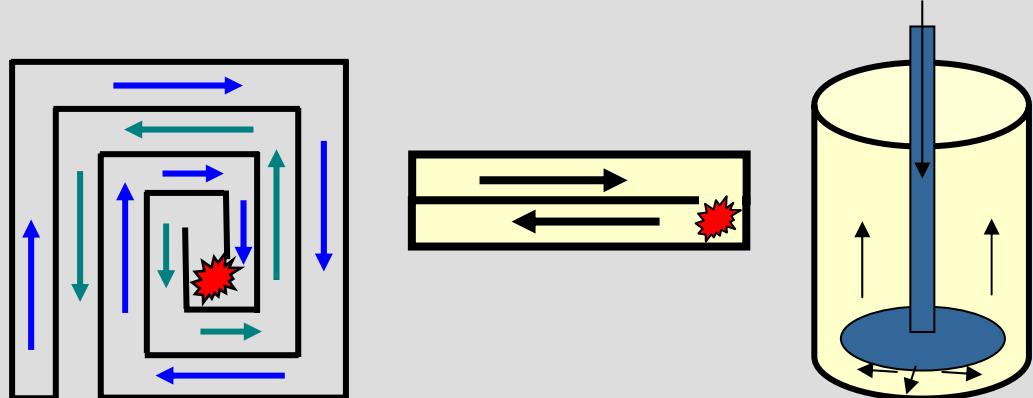
► Reactor

- Packed Bed
- Coated walls
- Engineered



► System

- Spiral – Swiss roll
- Counter-flow
- Radial



Material Selection

Material	Benefits	Challenges
Low Temp. Ceramic Casting	<ul style="list-style-type: none">- Flexible fabrication- Refractory and durable- Low costs- No clean room required	<ul style="list-style-type: none">- Non-standard fabrication- Low thermal conductivity- Sealing- Fragile
Silicon	<ul style="list-style-type: none">- Well characterized Si fab. techniques- High precision- Low to modest costs	<ul style="list-style-type: none">- Modest thermal conductivity- Fragile- Requires clean room
Metal	<ul style="list-style-type: none">- Conventional fabrication techniques- Durable- Low costs- No clean room required	<ul style="list-style-type: none">- High thermal conductivity- Poor compatibility with ceramics and glass

Adapted from Thompson, L. T. Microreaction Technology and Process Intensification Symposium at the 226th American Chemical Society National Meeting, New York, NY, 2003; p abstract 35.

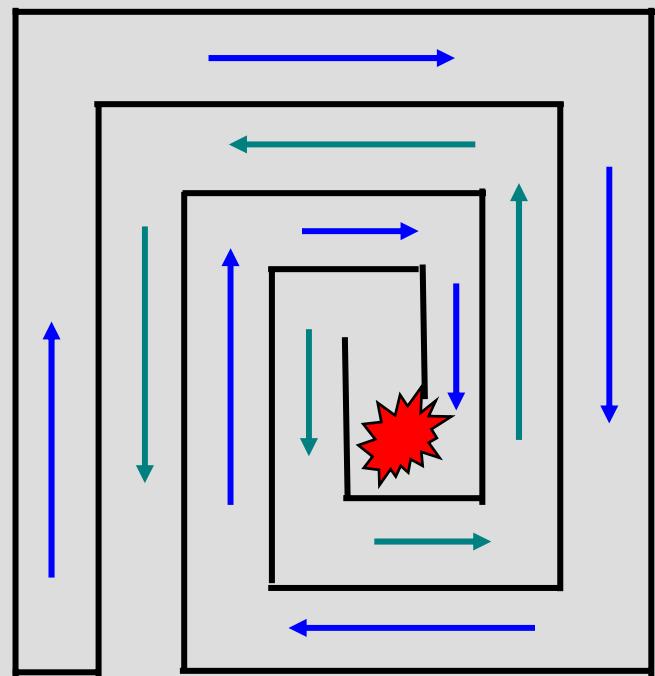
Material Properties

If thickness of SS is << zirconia, then the thermal conductivity can be less!

Material	Thermal Conductivity (W/m*K at 600K)	CTE (1/C) RT	Ductility
Silicon	62	2.6×10^{-6}	Brittle
SiO_x	5	5×10^{-7}	Brittle
Silicon nitride	11	3.6×10^{-6}	Brittle
Alumina	16	8.8×10^{-6}	Brittle
Zirconia	~3	10^{-6}	Brittle
316 SS	18	16×10^{-6}	Ductile

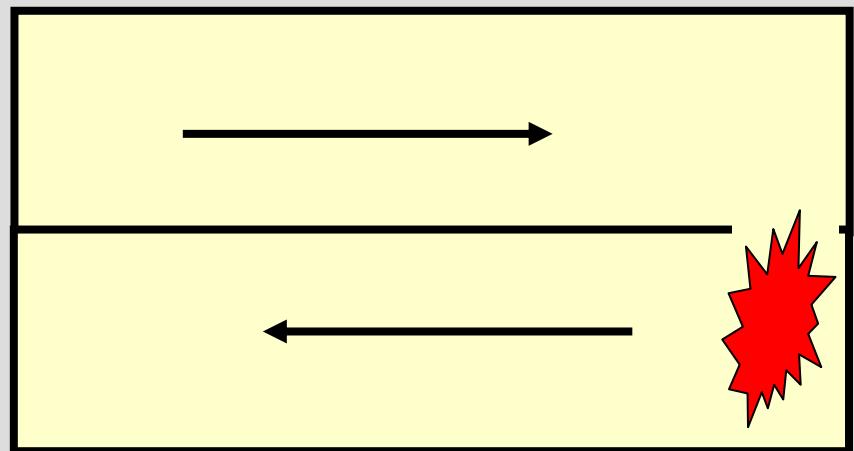
Design 1: Swiss Roll

- ▶ Reactor: Engineered Catalyst
- ▶ Materials: LTCC – low thermal conductivity
- ▶ Results: Never built
 - Device would be large
 - Complex
 - Thick walls required for strength
 - Sealing issues



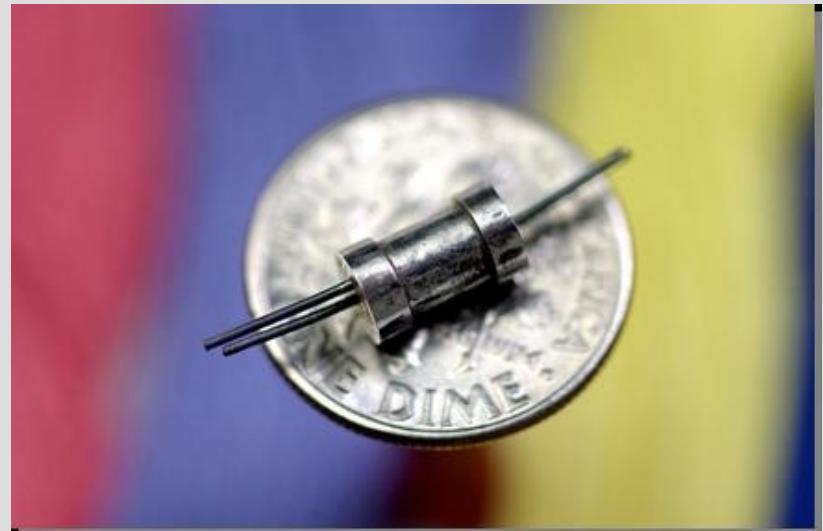
Design 2: Counter-Flow

- ▶ Reactor: Engineered Catalyst
- ▶ Materials: LTCC – low thermal conductivity
- ▶ Results: Built
 - Small device
 - Simple to build
 - Fragile – kept breaking
 - Sealing issues



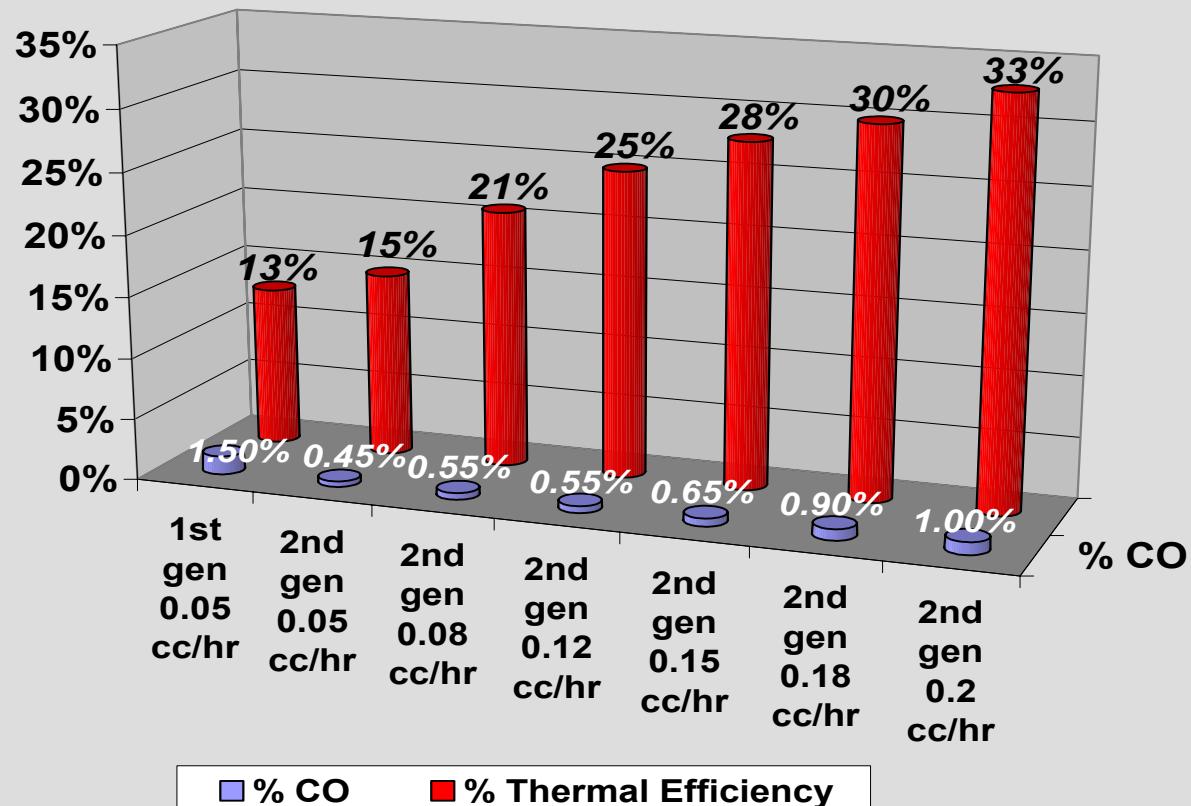
Design 3: Radial

- ▶ Reactor: Engineered Catalyst
- ▶ Materials: Stainless Steel – higher thermal conductivity but thinner walls
- ▶ Results: Built and tested
 - Surprisingly efficient, 9%
 - Small
 - Simple
 - Strong
 - No sealing issues
- ▶ Gas composition:
 - $H_2 = 72\text{-}74\%$
 - $CO_2 = 24\text{-}26\%$
 - $CO = 0.9\text{-}1.9\%$



Design 4: Radial

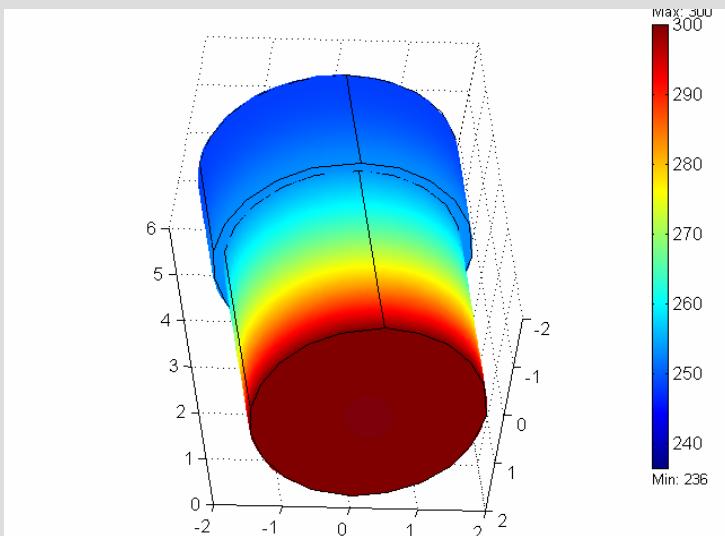
- ▶ Reactor: Mixed Engineered + Packed Bed
- ▶ Results: Improved efficiency with reduced CO⁻



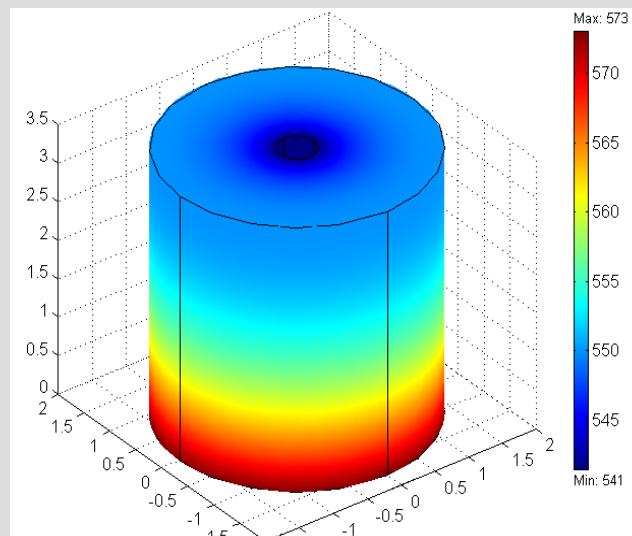
Design 4: Radial – Model

- ▶ Heat loss: conduction > radiation + convection
- ▶ Methanation reactor integration feasible
 - $T = 230\text{--}270^\circ\text{C}$

Surface temperature profile (C)
(with methanation reactor)

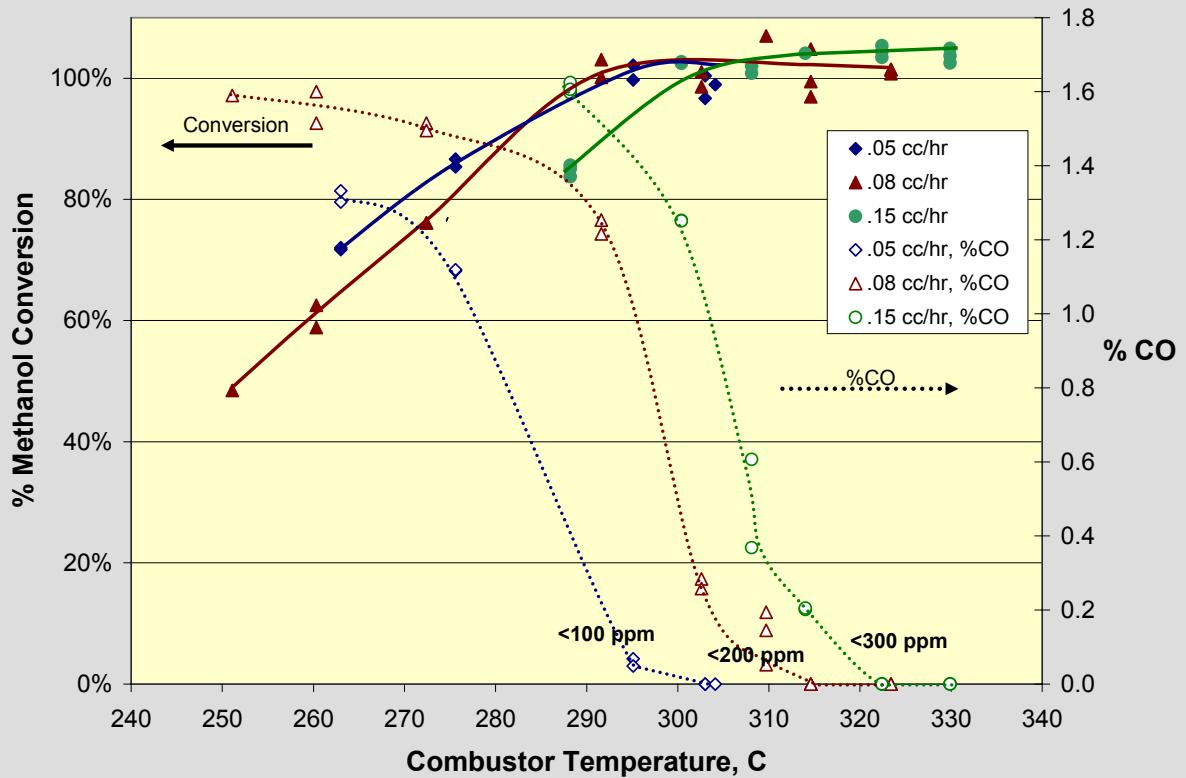


Surface temperature
profile (K)



Design 5: Radial

- ▶ Reactor: Mixed Engineered + Packed Bed
- ▶ Results: Improved efficiency with lower CO concentration



Radial Design Summary

Feed Flow, cc/hr	T, °C	Gas Flow, sccm	H ₂ , %	CO ₂ , %	CH ₄ , %	CO, ppm	Approx. Thermal Eff., %
0.05	304	1.0	69-70	25	5.0-5.5	<100	9
0.08	323	1.65	69-70	25	6.0-6.2	<200	14
0.1	330	2.1	69-70	25	6.0-6.2	<200	17
0.15	345	3.2	68-69	25-26	5.3-6.0	<300	19