

# *Integrated Fuel Processor Development*

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*Argonne Electrochemical Technology Program*

*Objective: Determine reactor design, operating conditions that can meet PNGVs FP targets*

- \* Study processing steps individually - effect of operating conditions, obtain kinetic parameters (100 W)
  - \* with different catalysts
  - \* obtain desirable temperature profile for WGS
- \* Study the component and thermal integration (5 kWe level)
  - \* identify concepts that improve performance
    - u feed location, mixing, reactor size, S-removal effectiveness
  - \* validate models

# *Project Timeline*

Jun. 2001: Transfer fuel processor integration technology

Oct. 2000: Transfer fuel processor technology to private developers

May 2000: Initiate collaboration with LANL to integrate PrOx

Apr. 2000: Completed evaluation of two PrOx catalysts in process train

Dec. 1999: Compare WGS catalysts in integrated unit (ANL, CuZnO)

Aug. 1999: Demonstrated 100 ppm CO from iso-octane, in process train

Apr. 1999: Operate integrated FP with gasoline - 30% H<sub>2</sub> in product

Mar. 1999: Started experiments with process train

Nov. 1998: ATR, sulfur trap, and shift reactor integrated design

Nov. 1997: Demonstrate new ATR reforming catalyst in 3" tubular reactor

Feb. 1996: Transfer methanol reforming to GM R&D

Aug. 1995: Demonstrate methanol reforming in 2" reactor - 50% H<sub>2</sub> in product

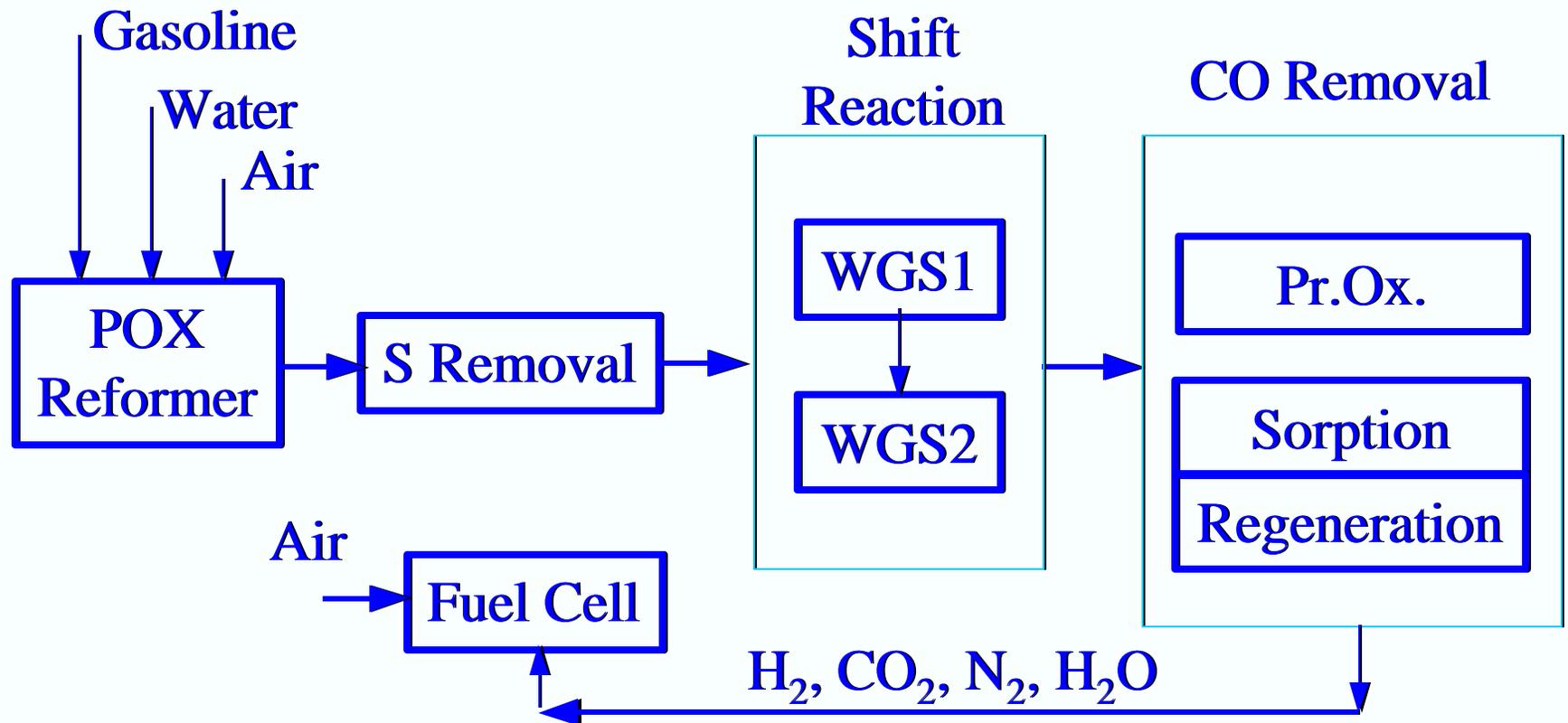
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<b><i>Milestone</i></b>	<b><i>Due</i></b>	
Process Train: 1% CO after WGS from gasoline reformat	Feb.	U
Integrated FP: 75% efficiency on gasoline	May	U
Integrated FP: 1% CO from gasoline	Jun.	U
Design optimum temperature profile for shift reactor	Jun.	U
Integrated FP: Measure start-up emission	Jun.	
Partnership with industrial developer for technology transfer	Jun.	U

# *Reviewers commented last year...*

- \* **Concentrate on fundamentals**
  - \* Catalyst, catalyst forms, operating conditions are being studied
- \* **Increase focus on integration, with modeling**
  - \* Modeling effort has been increased
  - \* Reactor configuration with model, followed by model validation
- \* **Work with system integrators and developers**
- \* **Address dynamic and idle conditions**
- \* **What is being learned?**

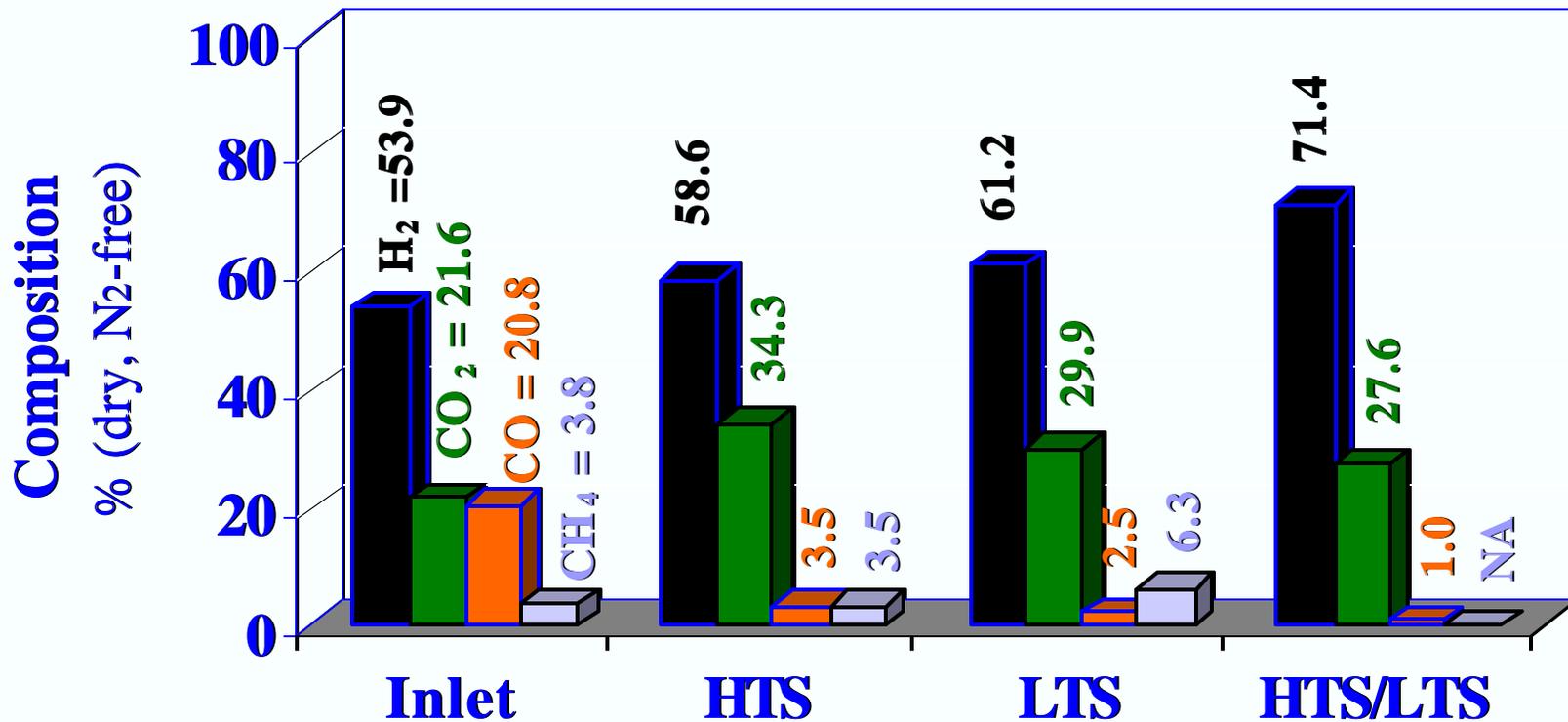
*The process train is used to study the components individually or in series*



\*The catalyst performance is verified under more realistic conditions in an engineering scale reactor.

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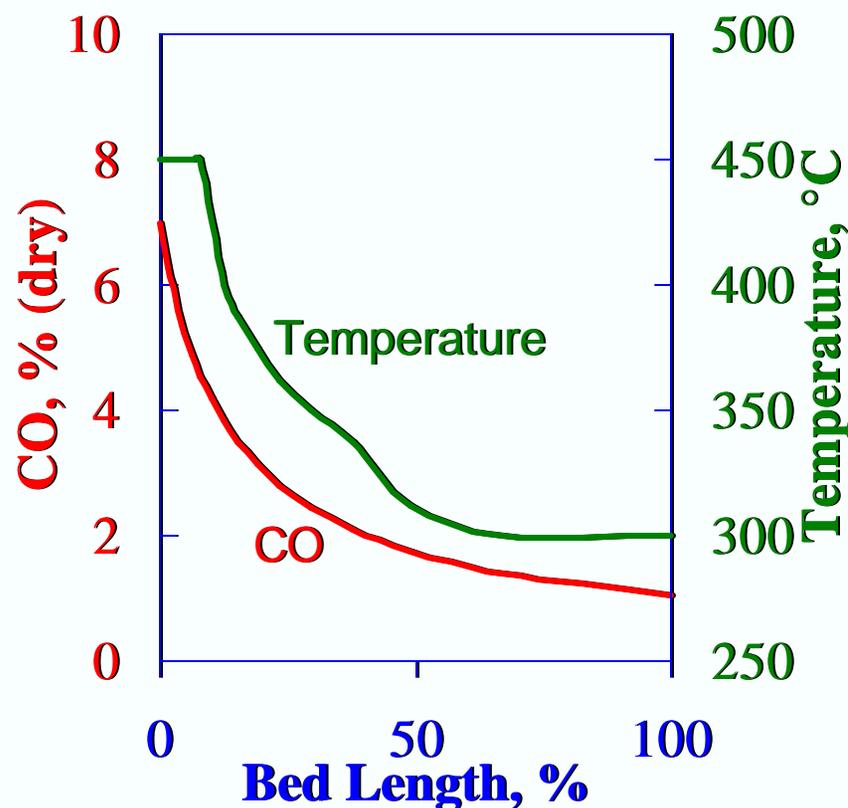
# *The shift reactors were configured to achieve 1% CO*



HTS: Fe/Cr Oxide, T = 300°C  
LTS: ANLWGS, T = 215°C

## *A declining temperature profile for the shift reactor was calculated*

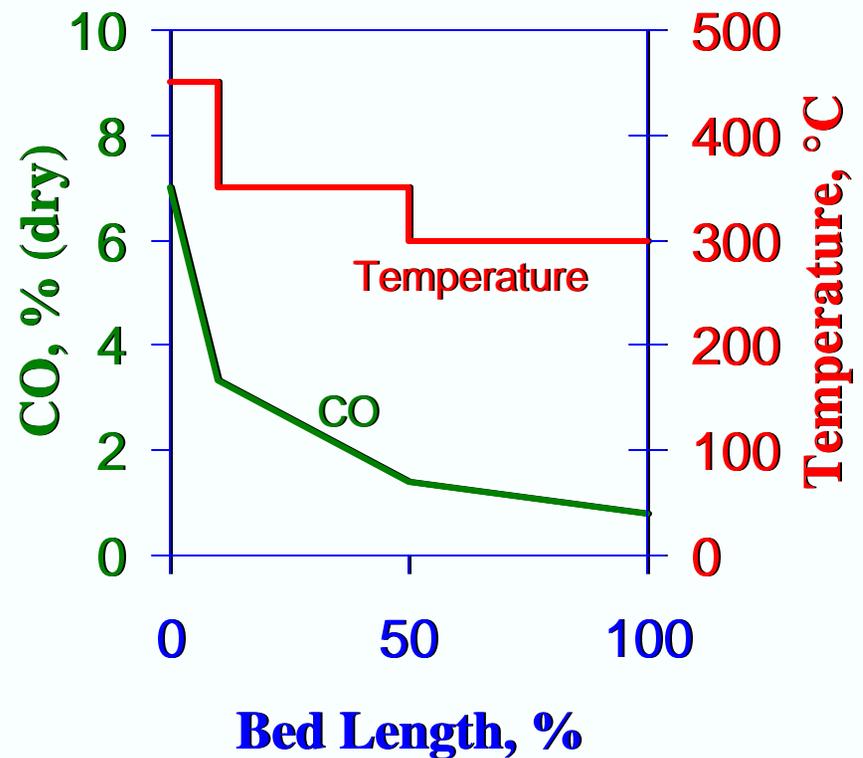
- \* For conversion of 85% of CO from ATR
- \*  $\text{CO}_{\text{in}} = 7\%$ ;  $\text{CO}_{\text{out}} = 1\%$ ;  
 $\text{H}_2\text{O}/\text{CO} = 3$
- \* Catalyst: ANL-WGS
- \* A shift reactor with the determined temperature profile will require the least catalyst



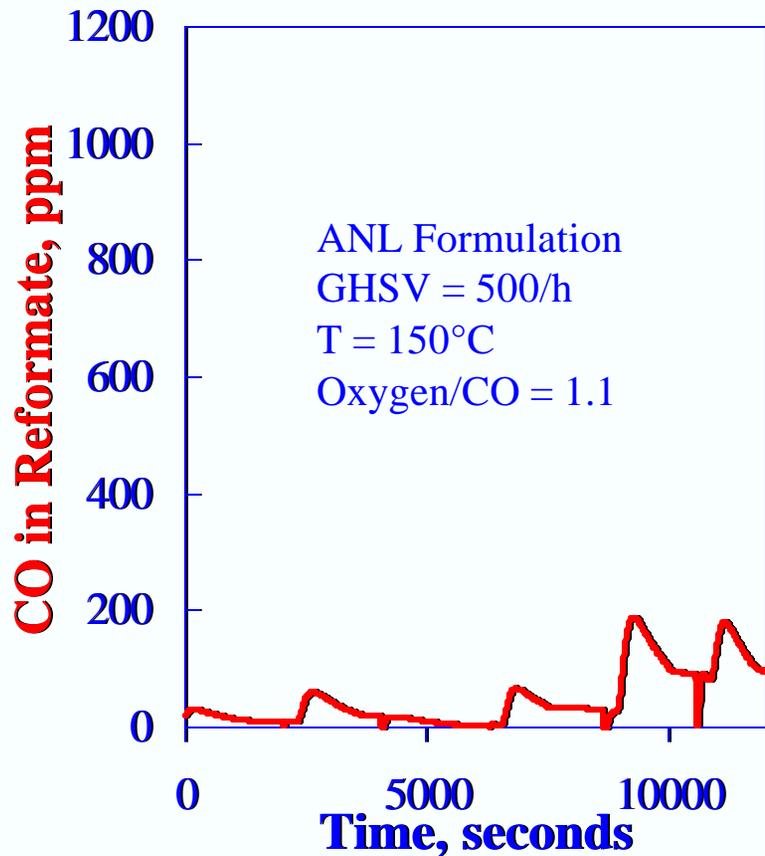
*Multiple isothermal beds can be used to approach optimum-temperatures reactor*

\* 3 isothermal reactors will require 25% longer bed

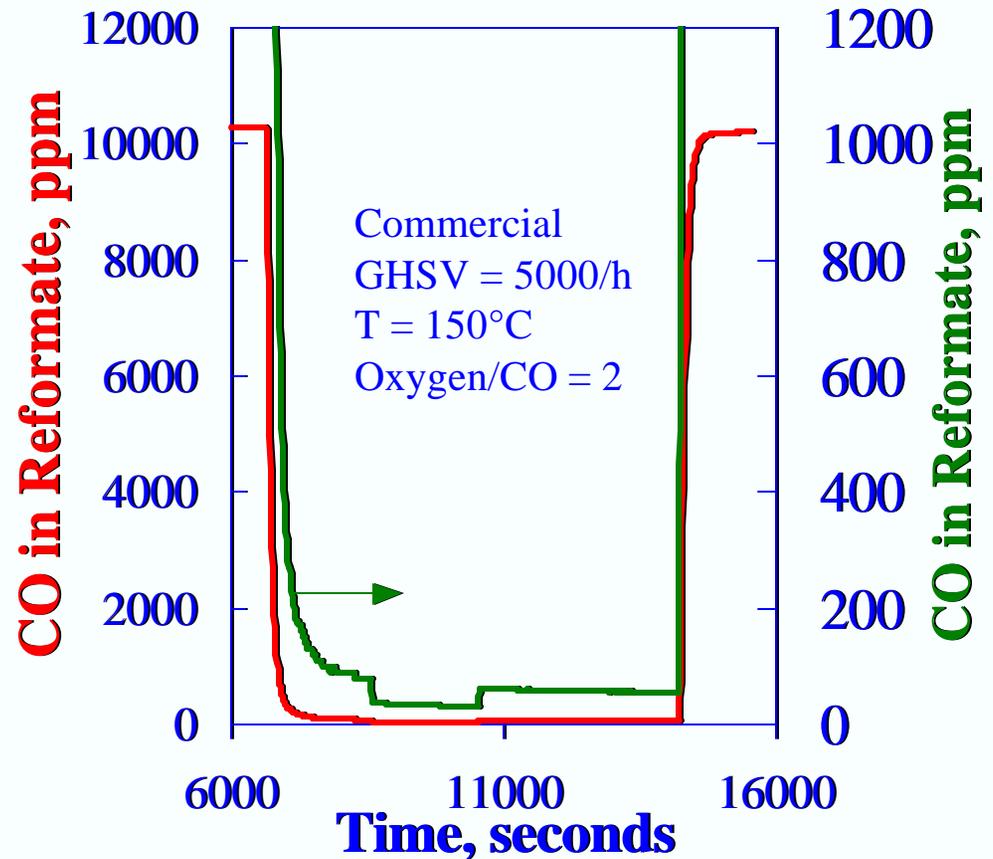
\* Multiple temperature zones can be approximated in an integrated reactor



# *Preferential oxidation catalysts are being evaluated*



\* Mn-based formulation was not stable



\* Methanation activity observed at higher temperatures

# *The engineering unit is used to verify integration concepts, catalyst performance*

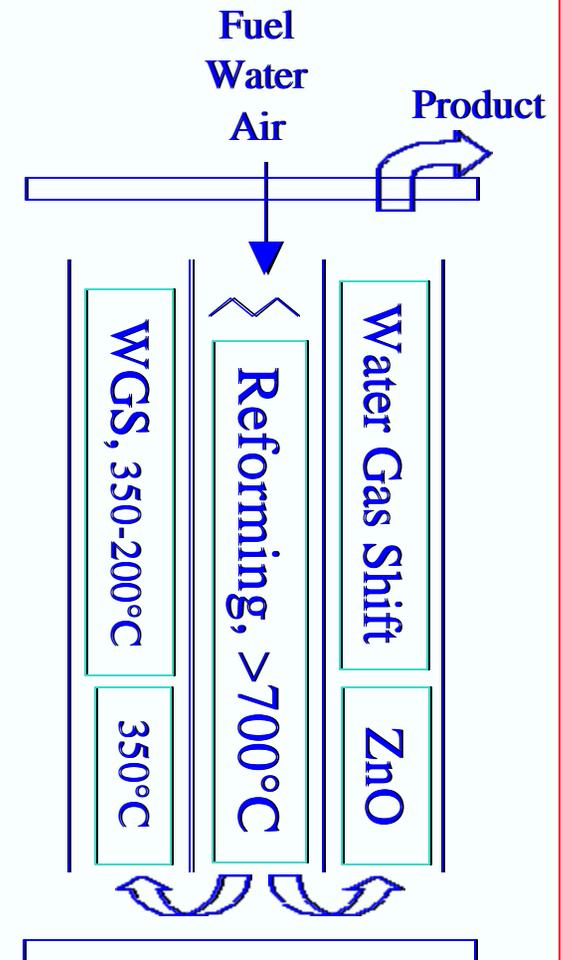
## \* Includes:

- \* fuel vaporizer
- \* steam generator
- \* air preheater
- \* autothermal reformer
- \* sulfur trap
- \* shift reactor

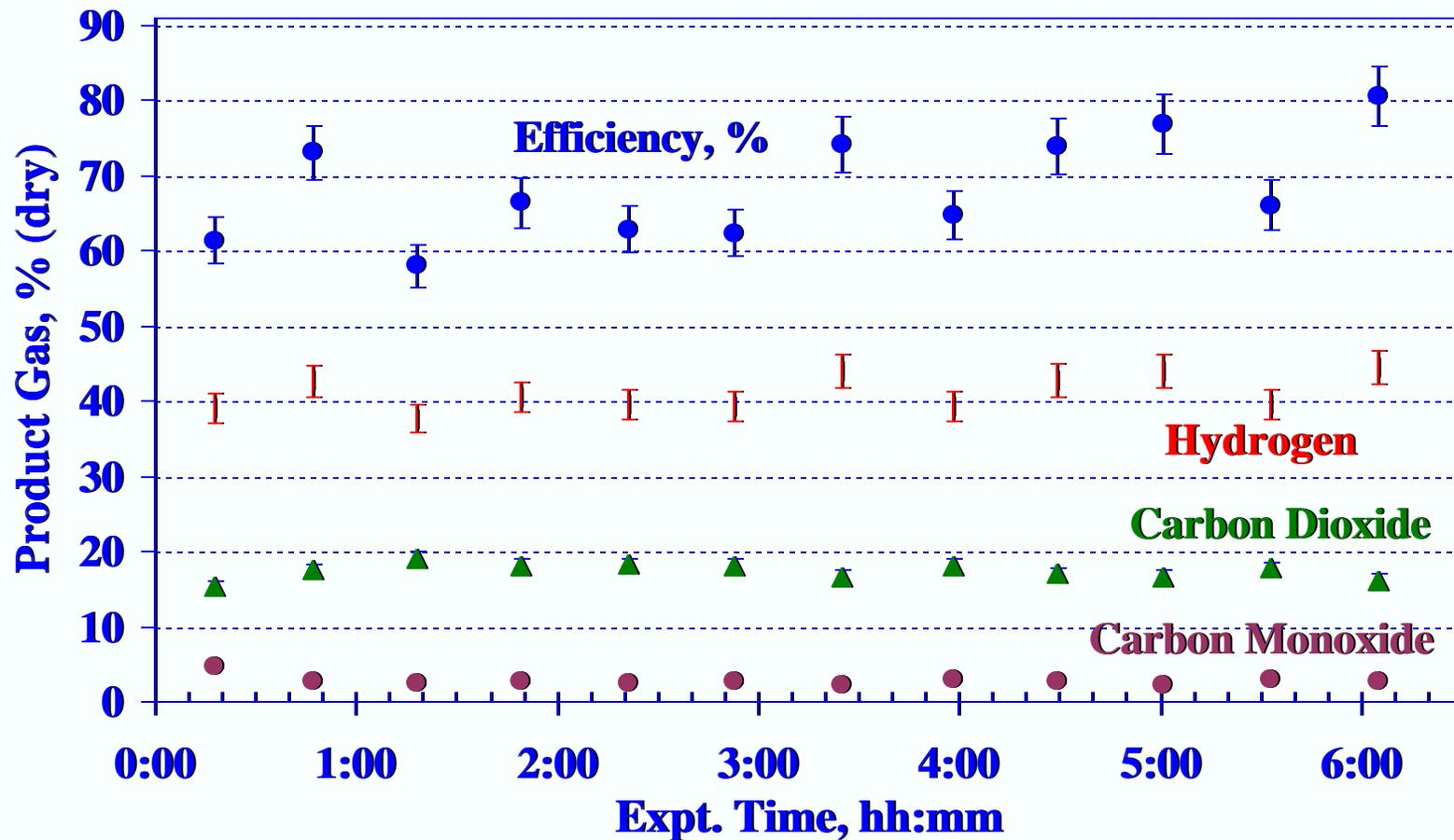
\* Capacity - 10 kWe

\* Fuel: gasoline, alcohol

\* 7-in. dia; 12-in long, 7-Liter



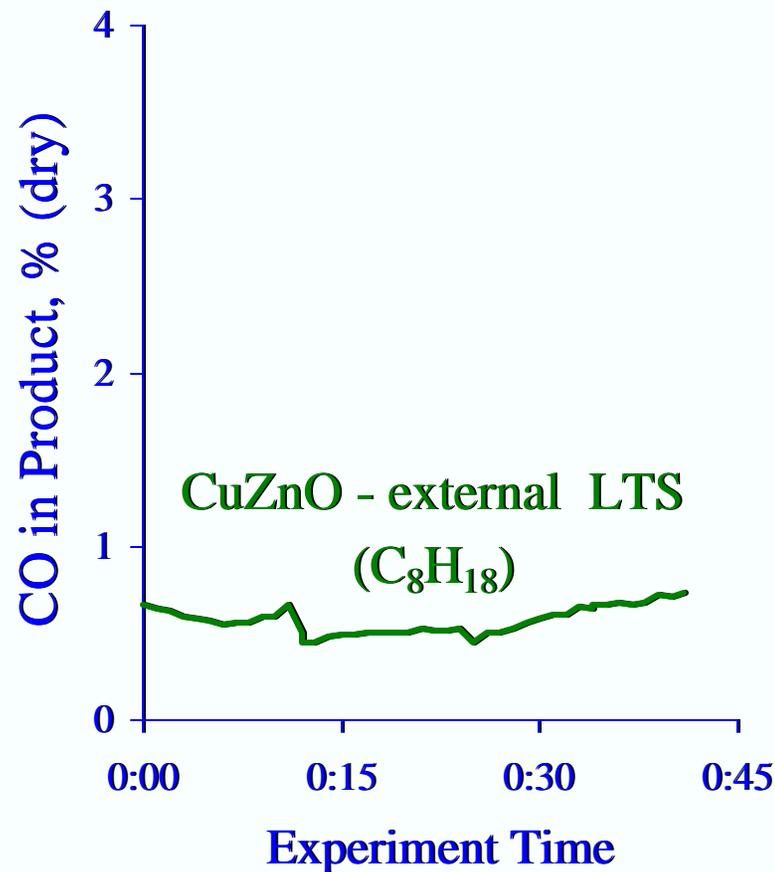
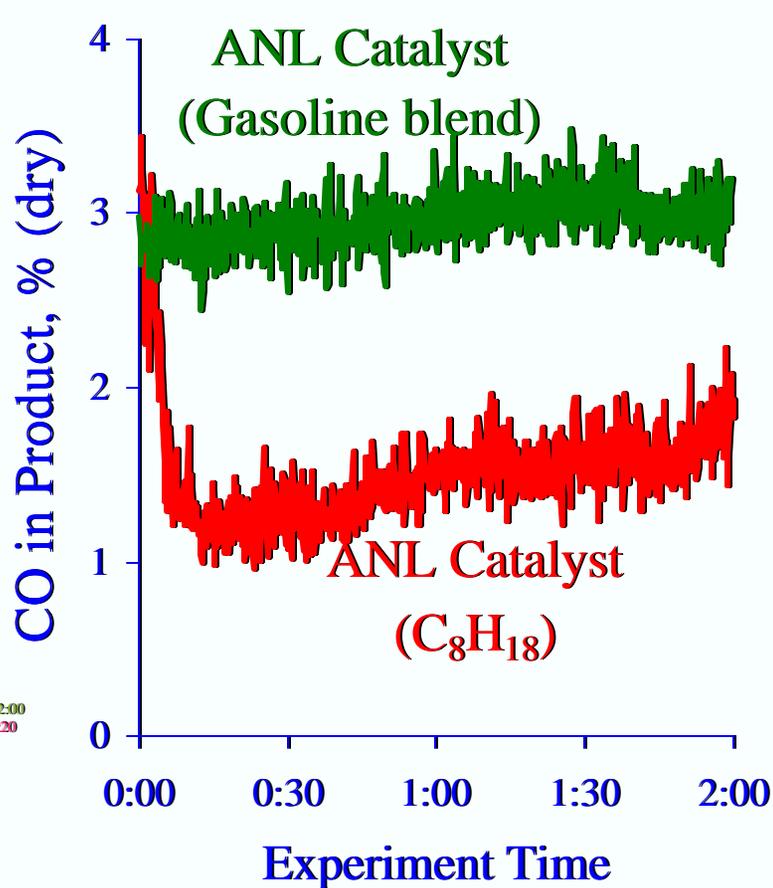
*The processor achieved 88% of theoretical efficiencies, with 42% hydrogen*



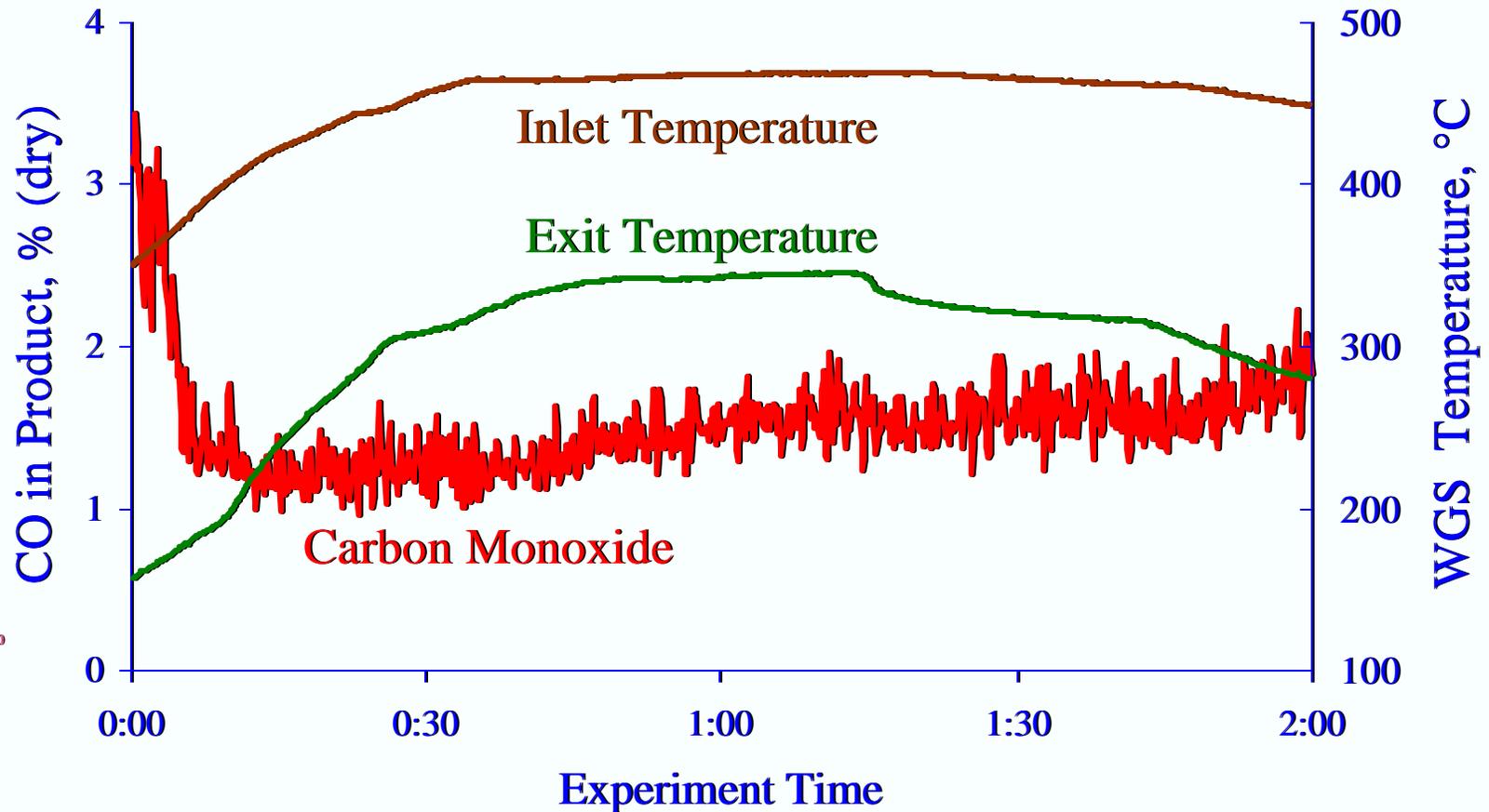
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*Less than 1% CO was achieved,  
but difficult to maintain*

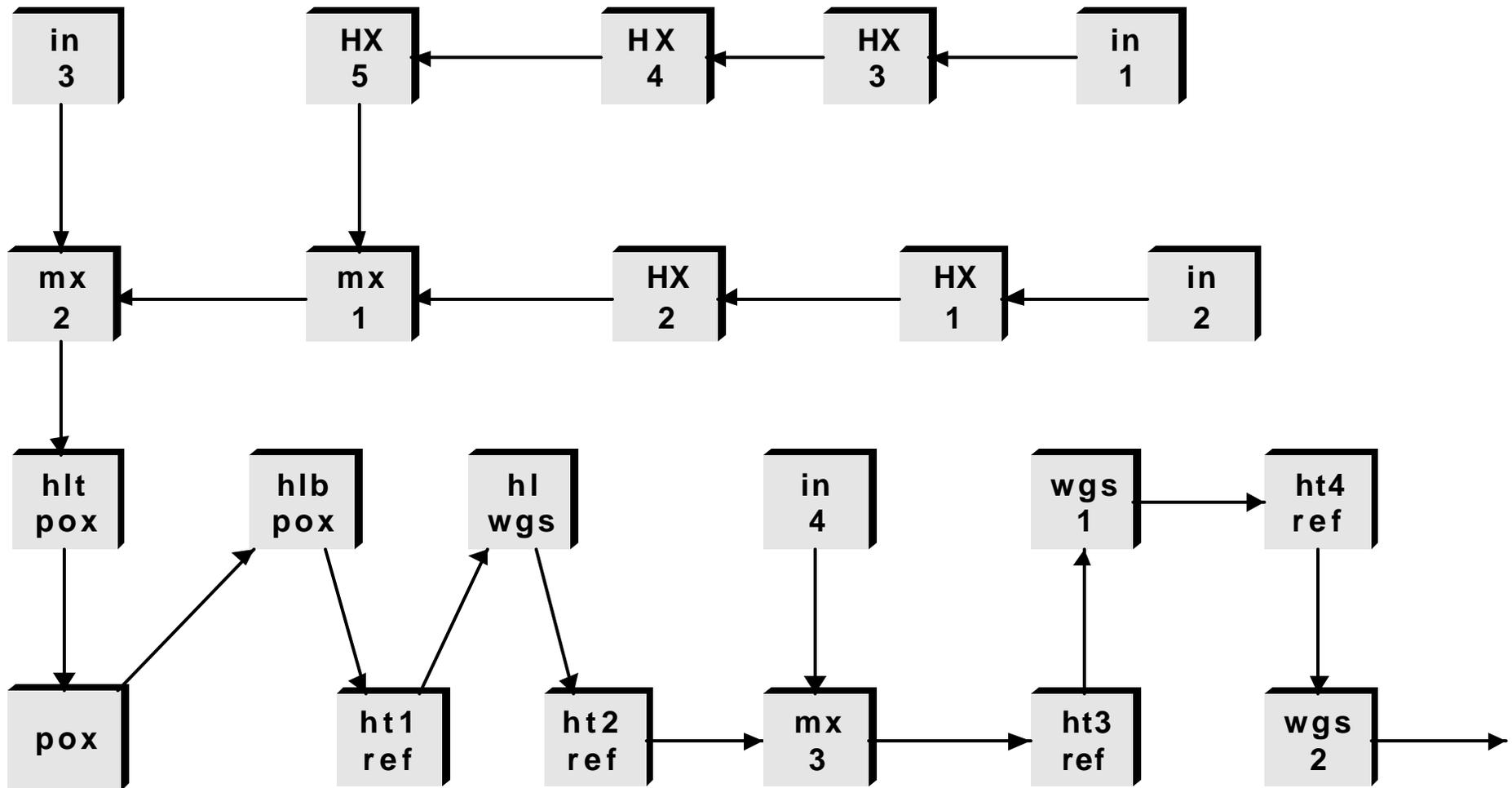


# *CO concentration is strongly dependent on WGS temperature*

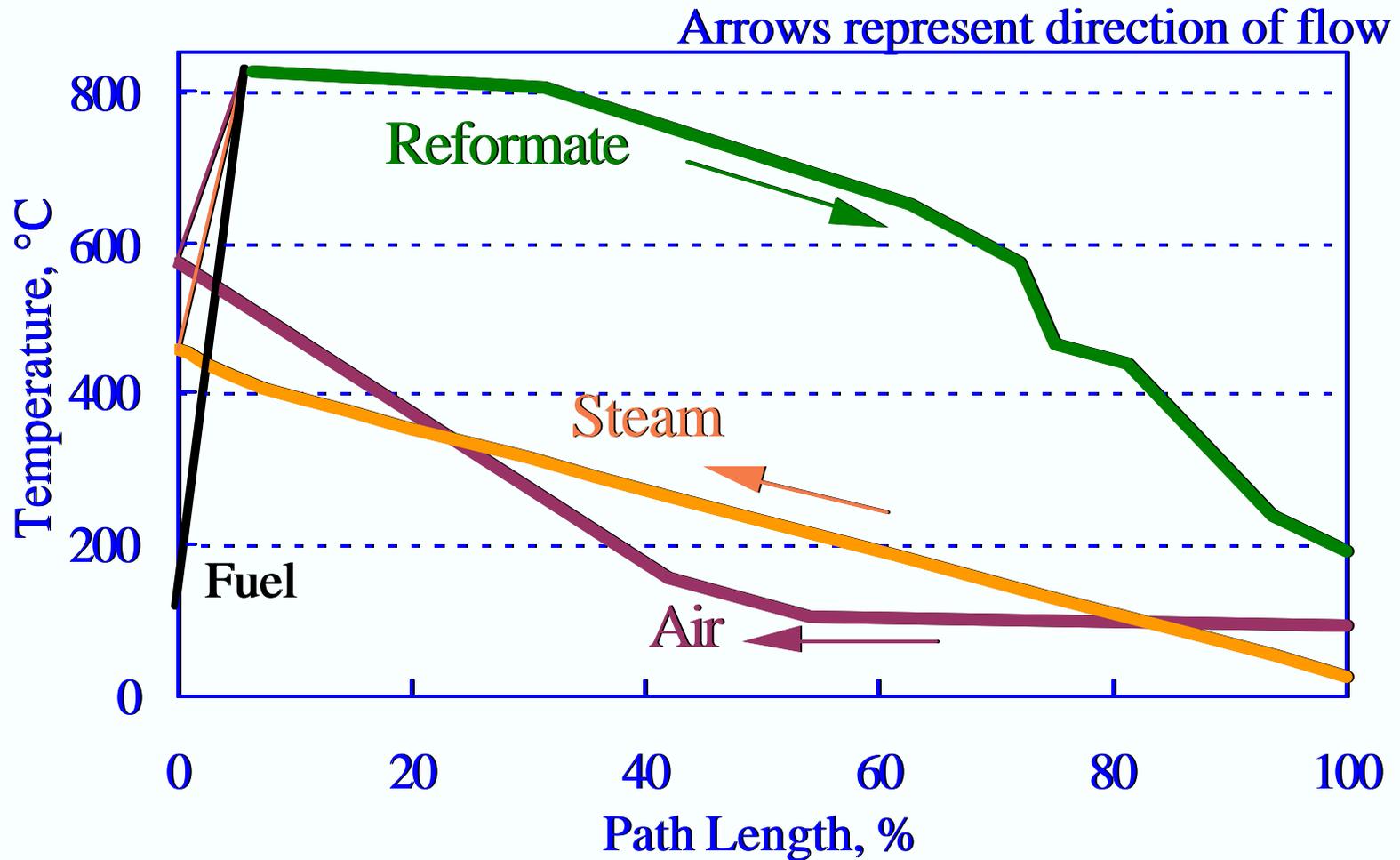


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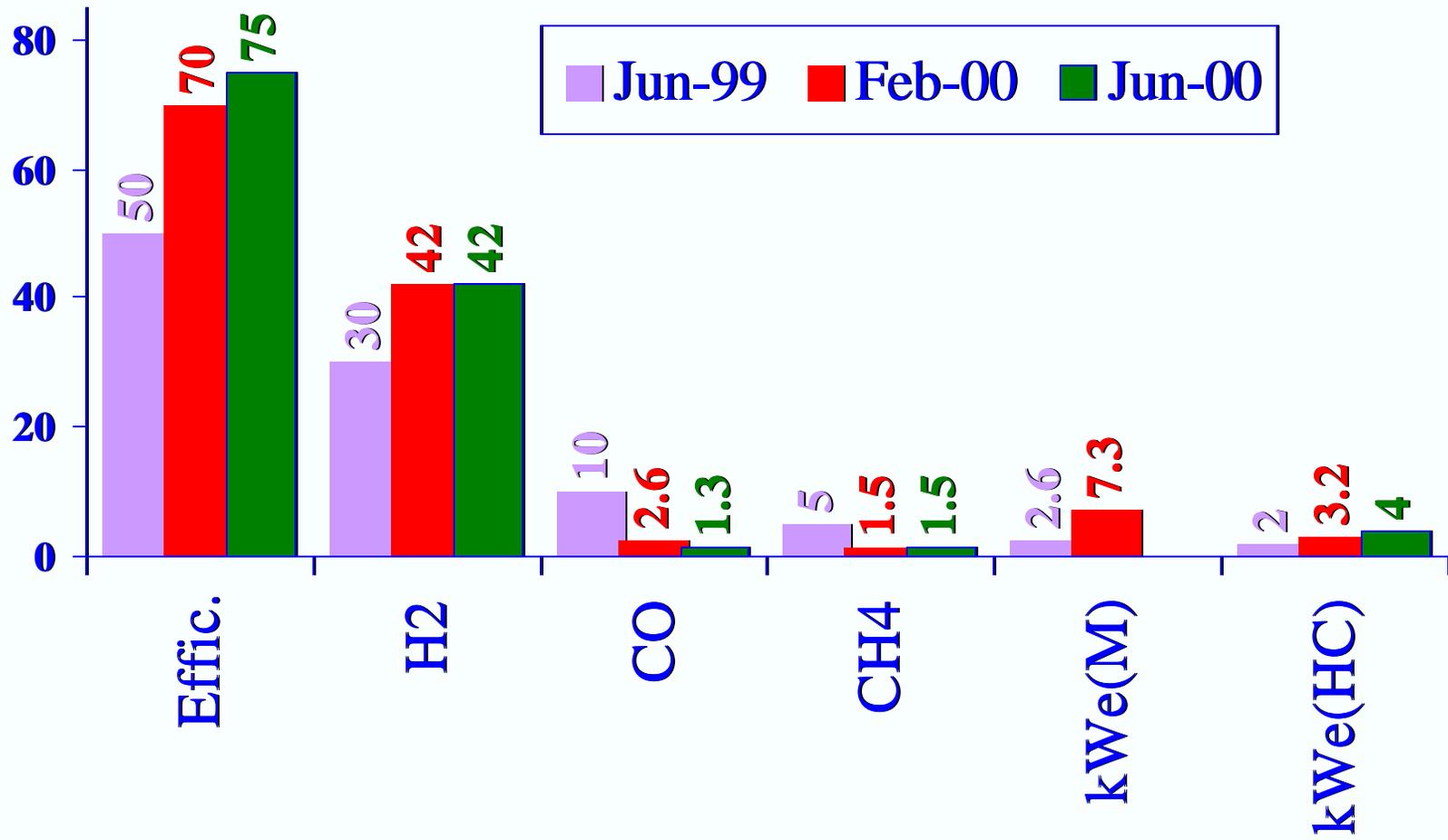
# *Effectiveness of reactor designs are determined with GCTool models*



# *Reactant streams are used to control reformat gas temperature*



# *Addressing fundamentals of integration has led to significant performance gains*



## *Technology Transfer, External Interactions*

- \* Autothermal reforming catalyst technology licensed to United Catalysts
- \* ANL will help a major system developer with fuel processor technology
  - \* for an automotive fuel cell system
- \* ANL will design a fuel processor for a start-up company (Unitel Technologies)
- \* Collaborating with LANL to integrate PrOx into ANL's integrated design

# *Future Plans*

- \* Determine effect of operating conditions on size and performance on all processing steps
- \* Investigate engineering designs that reduce rapid start-up time
- \* Integrate PrOx with LANL collaboration
- \* Transfer technology to private enterprise

# *What have we learned?*

- \* ANL shift catalyst performance was verified in the process train and the integrated unit
  - \* low temperature activity needs to be improved
- \* Next to catalyst activity, temperature control is the most important for the shift reactor
  - \* idealized temperature profiles can be calculated
  - \* integrated reactors can provide declining temperature profile (discretization possible)

*We have verified catalyst performance, and learned integration opportunities through modeling and hardware testing*

- \* **Non-noble metal PrOx catalysts are feasible**
  - \* need increased activity, durability
- \* **Low temperature PrOx catalysts are more selective (for CO), prone to methanation**
- \* **Fuel processor efficiencies can approach theoretical limits with proper integration**