

PNNL-35030	
	Investigating carbon stabilization in soils via mineral adsorption
	What are we missing when we define mineral associated organic matter.
	September 2023
	Qian Zhao Odeta Qafoku Kenton Rod
	U.S. DEPARTMENT OF <b>ENERGY</b> Prepared for the U.S. Department of Energy under Contract DE-AC05-76RL01830

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# Investigating carbon stabilization in soils via mineral adsorption

What are we missing when we define mineral associated organic matter

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Pacific Northwest National Laboratory Richland, Washington 99354

#### Abstract

Understanding carbon (C) fluxes released from soil to atmosphere is critical to regulating global climate change. Soil minerals play a crucial role in stabilizing C, based on recent studies which have found that mineral associated C can be stored in soils for decades to centuries longer than non-mineral associated C. We aim to maximize C sequestration via mineral adsorption in soils. To achieve this objective, it is important to understand the potential impacts of C stabilization via adsorption to mineral surfaces. Particularly, investigating the quantitative impacts of C sorption to minerals is novel to understanding and implementing multi-scale biogeochemical processes of C stabilization in soils. We will conduct a literature review on potential impacts of C sorption to minerals, thereby contributing to the net C storage in soils. Our findings will allow us to quantitatively understand soil C changes and durability at molecular and ecosystem scales, fulfilling existing scientific gaps in the community as well as interests of sponsors, such as DOE-BER.

#### **Summary**

This project provides insights into the role of minerals on soil carbon (C) storage, sequestration, and contribution to the global C cycle. The outcomes of this study demonstrated how mineral adsorption processes might enhance C storage in terrestrial ecosystems. We have conducted an outline of a manuscript that focuses on the diverge quantification methods of mineral associated organic matter (MAOM) pool in soil could lead to an uncertain estimation of global mineral associated C stock in models. Meanwhile, we also build up a body of literatures as well as a dataset of MAOM-C stock and the fraction of MAOM-C to total soil organic carbon (SOC) across different ecosystems from literatures. Along with MAOM pool size data, this dataset also includes site metadata, including mean annual temperature (MAT), mean annual precipitation (MAP), soil moisture, soil pH, etc. We are continuously building this dataset by synthesize data from literatures. Information from this LDRD project will be used to develop a literature review manuscript and potential DOE BER proposals.

#### **Acknowledgments**

This research was supported by the **EBSD Mission Seed**, under the Laboratory Directed Research and Development (LDRD) Program at Pacific Northwest National Laboratory (PNNL). PNNL is a multi-program national laboratory operated for the U.S. Department of Energy (DOE) by Battelle Memorial Institute under Contract No. DE-AC05-76RL01830.

# **Acronyms and Abbreviations**

soil organic matter (SOM) mineral associated organic matter (MAOM) carbon (C) soil organic carbon (SOC) mean annual temperature (MAT) mean annual precipitation (MAP)

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#### **1.0 Introduction**

Understanding carbon (C) fluxes released from soil to atmosphere is critical to regulating global climate change. Soil minerals play a crucial role in stabilizing C, based on recent studies which have found that mineral associated C can be stored in soils for decades to centuries longer than non-mineral associated C (Lützow et al 2006, Mikutta et al. 2019, Schmidt et al. 2011). For example, microbial-derived C can be stabilized in soil over years through adsorption to amorphous Fe hydroxides (e.g., ferrihydrite), Al oxides, and Ca-bearing minerals (e.g., calcite), due to their high surface areas, the prevalence of surface hydroxyl groups, and bridging capacities. This project probes the impact of organo-mineral and microbe-mineral interactions on C assimilation and protection in soil to guide soil C sequestration. Thus, our findings allow us to quantitatively understand the persistent soil C that associates with soil minerals at molecular and ecosystem scales, fulfilling existing scientific gaps in the community as well as interests of sponsors, such as the U.S. Department of Energy (DOE)-Biological and Environmental Research (BER). This effort aligns well with the directorate objectives of PNNL Earth and Biological Sciences Directorate (EBSD) on Durable Soil Carbon Storage for Atmospheric Carbon Dioxide Removal.

#### 2.0 Current understanding of C budget in the MAOM pools

Second State of the Carbon Cycle Report (SOCCR2) has identified the soil C stock in top 1m soil from North American is 400 PgC (Lajtha et al. 2018). Estimates of global soil organic carbon (SOC) stocks vary from 684 to 724 Pg in the surface 0.3 m (Baties 1996), the Intergovernmental Panel of Climate Change (IPCC) standard sampling depth (Aalde et al. 2006), and (Sokol et al., 2019)1462 to 1548 Pg in the top meter and 2,060 ± 220 Pg C to 2 m (Batjes 1996). Diverse C inputs result in heterogeneity of the SOM pool. Studies showed both microbial-derived C and plant-derived C are major sources of mineral-associated organi matter (MAOM). The formation and accumulation of SOC are primarily derived from plant- and microbial C (Angst et al., 2021; Ma et al., 2018; Sokol & Bradford, 2019). While the significance of plant residues to SOM persistence is well recognized due to the slow cycling of plant debris (e.g., lignin and phenols) (Angst et al., 2021; Schmidt et al., 2011), the importance of microbial products and residues (hereafter 'microbial necromass') in the slowly cycling SOC pool has only recently been appreciated (Fan et al., 2021; Kallenbach et al., 2016; Liang et al., 2019; Wang et al., 2021; Wu et al., 2023). Microbial necromass contributes to 33-62% of total SOC in various types of ecosystems (measured as amino sugars) (Liang et al., 2019; Wang et al., 2021). However, the long-term persistence of microbial necromass in the SOM pool depends on the associations with soil minerals (Kästner et al. 2021), even though the specific necromass-mineral interactive mechanisms are still unclear.

The estimations of MAOM pool in soils have been largely studied in the last decade. Studies showed the proportion of MAOM to total SOM varies significantly, ranging from 25%-90% (Table 1). The proportion of the MAOM pool varies across different ecosystem types with 60% in forest, 72% in grassland, and 79% in cropland. Lugato et al (2021) did size fractionation on 400 soils and quantified MAOM with a machine learning (ML) approach. The model was then used to predict the C and N in the MAOM fraction for the 9,229. The average mineral associated C stocks in European grasslands and forest soils varied between 15 and 38 g C kg<sup>-1</sup> soil (Figure 1).

Site or ecosystem type	Proportion of MAOM to total SOM	Methods	Reference
14 Forest soils in the USA	0.6-57.8% (average 37.8%)	Dithionite extraction	<u>Zhao et al.</u> 2016
4 wetland soils in China	7-91%	Size and density	<u>Liu et al.</u> 2023
Mangrove soils in the Philippines	~15 %	Dithionite extraction	<u>Dicen et</u> al., 2019
Arable soils in China	6.2–31.2 % Agriculture soils – grassland and arid	Dithionite extraction	<u>Wan et al.,</u> <u>2019</u>
10 Peatlands in China	1.64–5.94 %	Dithionite extraction	<u>Huang et</u> al., 2021a
186 grassland and forest soils in European	25-85% (60% in forest, 72% in grassland, 79% in cropland)	Size	<u>Cotrufo et</u> <u>al., 2019</u>
1451 all ecosystem types	65%	Size	<u>Georgiou</u> <u>et al. 2022</u>

#### Table 1. Percentage of Fe-OC to total OC in natural soil and sediment environments.

#### Grasslands 15.8 ± 12.0 % Dithionite Tibetan alpine grasslands in Fang et China extraction al., 2019 Meadow soils in the Qinghai-4.1-25.6 % Dithionite Mu et al.. Tibetan Plateau 2020 extraction **Sediments** Marine sediments in Mexican and $21.5 \pm 8.6$ % Dithionite Lalonde et Indian margins, the Southern extraction <u>al., 2012</u> Ocean, the St. Lawrence estuary and gulf, and the Black Sea East China Sea sediments 2.77-31.5 % Dithionite Ma et al., extraction 2018 Changiang estuary sediments in 7.4 ± 3.5 % Dithionite Zhao et China extraction al., 2018 Wax Lake Delta sediments in the $\sim 15.0$ % Dithionite Shields et USA al., 2016 extraction Sediments in Eurasian Arctic 0.5-22% Dithionite Salvadó et Shelf extraction <u>al., 2015</u> Saanich, Arabian Sea, Mexican 25.7-62.6 % Dithionite Barber et margin, and St. Lawrence estuary extraction al., 2017 Permafrost Permafrost soils in northern 13.68±2.31 % Dithionite Joss et al.. Alaska extraction 2022 Permafrost soils in the Qinghai-19.5 ± 12.3 % Dithionite Mu et al., **Tibetan Plateau** extraction 2016 Discontinuous permafrost region 9.9-14.8% Dithionite Patzner et in Sweden extraction al., 2020



Figure 1. **a**–**c**, Carbon content (gC kg<sup>-1</sup> soil) in POM (**a**) and MAOM (**b**), and ratio between MAOM carbon and total SOC (**c**) from Lugato et al (2021).

#### 3.0 Different parameters impact on MAOM pool size

#### 3.1 Quantification methods

There are diverse methodologies for quantifying the MAOM pool in soil. Abramoff et al (2021) compiled 402 laboratory sorption experiments and quantified, for the first time, the sorption capacity of mineral soils to DOC for six soil orders. They find that mid- and low-latitude soils and subsoils have a greater capacity to store DOC by sorption compared to high-latitude soils and topsoils. The global additional DOC sorption is estimated to be 107  $\pm$  13 Pg C to 1 m depth, projecting a 7% increase in the existing total carbon stock.

Another two major MAOM quantification methods are chemical extraction and size fractionation. Chemical extraction includes sodium pyrophosphate, sodium dithionite, oxalate acid, and HCI. These solvents target to dissolve specific minerals from soils, thereby releasing OC that associates with these minerals. Among these solvents, sodium dithionite has been widely used to quantify mineral bound OC in addition to quantify extracted Fe, Al, and other metals (Lalonde et al 2012, Zhao et al. 2016). Meanwhile, size fractionation separates the MAOM pool by less than 53µm particle size. This approach is easy and high throughput so that can be applied to a large number of sample sites. Dithionite extraction could underestimate the MAOM pool as it only extracts a portion of minerals from soils but not all, whereas size fractionation most likely overestimates the pool as there could be non-mineral bound C within particles less than 53µm. These different quantification approaches result in large variations in estimating the MAOM pool size in soils. For instance, dithionite extraction estimates about 25% of total SOC as MAOM C, whereas size fractionation estimates an average of 65% of total SOC as MAOM C pool. Such different estimations on MAOM pool result in diverge projections in continental scale MAOM pool. Given the total C stock, we estimated 60-260PgC in the MAOM pool in top 1m soil in North America. This diverge estimations on the MAOM pool size will impact on the accuracy of C cycling predictions by biogeochemical process-based models.

#### 3.2 Other factors

MAOC to reactive Fe ratios (OC:Fe) can indicate the mechanism of Fe-OC interactions. A mass ratio over 0.22 is indicative for Fe-OC associations predominantly formed by co-precipitation or chelation (Wagai and Mayer, 2007). Below an OC:Fe mass ratio of 0.22, OC is mainly assumed to be sorbed onto Fe minerals. Around 86% of the permafrost samples exceed an OC:Fe mass ratios of 0.22 (Joss et al. 2022). Other soil parameters, such as the bulk density, pH, and moisture content, were correlated with MAOM-C (Mu et al. 2022). Soil with higher pH results in higher fraction of MAOM-C, whereas higher soil moisture contents result in lower MAOM-C due to high microbial activities and consumption of available C.

#### 4.0 Implications and discussion on C cycling modeling

Ungeneralizable MAOM fractionation approach results in a large uncertainty to the estimation of MAOM pool, thereby impacting the estimation of C persistence in soil. Consensus on the experimental method of defining MAOM pool is necessary to the soil community. The lab-scale quantification of C storage via minerals is applicable to ecosystem scale and global scale. Soil mineralogy is a major factor to predict ecosystem-C behavior in terms of microbe-mineral interactions and nutrient-mineral interactions. The quantitative data from this project can be incorporated into process-based models, such as Earth and Environmental Systems (ESS), and/or artificial intelligence and machine learning (AI/ML) based models to improve model prediction on C storage by considering organo-mineral protection.

#### 5.0 Conclusion

This project provides insights into the role of minerals on soil carbon (C) storage, sequestration, and contribution to the global C cycle. The outcomes of this study demonstrated how mineral adsorption processes might enhance C storage in terrestrial ecosystems. We have conducted an outline of a manuscript that focuses on the diverge quantification methods of MAOM pool in soil could lead to an uncertain estimation of global mineral associated C stock in models. Meanwhile, we also build up a body of literatures (Figure 2) as well as a dataset of MAOM-C stock and the fraction of MAOM-C to total SOC across different ecosystems from literatures (Appendix A). Along with MAOM pool size data, this dataset also includes site metadata, including MAT, MAP, soil moisture, soil pH, etc. We are continuously building this dataset by synthesize data from literatures. Information from this LDRD project will be used to develop a literature review manuscript and potential DOE BER proposals.

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Figure 2. The list of literatures of quantifying mineral-associated OC in soil.

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# Appendix A – A working dataset of MAOM-C

Author '	Year	Lat	Lon	MAT	MA	Р	Soil.orde	Mineral Top.	lep Botto	m. Ve	egetat	frc_metl Clay	Silt	Sand	SiltClay	P Bulk.C	SiltClayC	POM_C	source	MAOM_	f MAOM
Amelung	1998	52.9	2 -105.0	8 0	9	456	Mollisols	HM	0	10 Gr	assland	Particle_Size	29 2	2 4	9 5	1 58.2	39.88	13.1168	Synthesis	45.0832	0.7746254
Amelung	1998	50.1	7 -107.	5 3	.0	343	Mollisols	HM	0	10 Gr	assiand	Particle_Size	22 2	27 5	1 4	9 42.3	27.03	7.5469	Synthesis	34.7531	0.8215863
Amelung	1998	45.3	-95.5	5 6	.1	565	Mollisols	HM	0	10 Gr	assland	Particle_Size	34 3	31 3	5 6	5 64	49.23	10.9116	Synthesis	53.0884	0.8295063
Amelung	1998	46.	5 -100.54	4	5	419	Mollisols Mollisols	HM	0	10 Gr	assland	Particle_Size	21 2	20 5	9 4	1 31.6 4 16.4	22.64	7.8396	Synthesis	23.7604	0.7519114
Amelung	1998	44.	5 -105.5	1 7	2	400	Aridisols	HM	0	10 Gr	assland	Particle_Size	24 3	14 4	2 5	8 18.6	12.57	3.3882	Synthesis	15.2118	0.8178387
Amelung	1998	41.2	2 -104.6	3 8	9	400	Mollisols	HM	0	10 Gr	assland	Particle_Size	14 1	2 7	4 2	6 8.1	6.99	1.7353	Synthesis	6.3647	0.7857654
Amelung	1998	40.	1 -103.1	3 10	.8	375	Mollisols	HM	0	10 Gr	assiand	Particle_Size	18 1	15 6	7 3	3 16.9	9.92	5.2655	Synthesis	11.6345	0.688432
Amelung	1998	40.2	5 -99.2	2 11	.6	666	Mollisols	HM	0	10 Gr	assland	Particle_Size	24 2	3 5	3 4	7 29.9	23.12	9.3886	Synthesis	20.5114	0.686
Amelung	1998	40.4	s -96.42	2 12	9	792	Mollisols	HM	0	10 Gr	assiand	Particle_Size	32 2	18 4	0 6	0 40.5	29.39	7.785	Synthesis	32.715	0.8077778
Amelung	1998	39.1	-96.3	5 12	.4	791	Mollisols	HM	0	10 Gr	assland	Particle_Size	33 2	17 4	0 6	0 30.7	24.24	8.6536	Synthesis	22.0464	0.7181238
Amelung	1998	37.	2 -95.10	6 14 8 17	1	1000	Alfisols Alfisols	HM	0	10 Gr	assland	Particle_Size Particle Size	18 3	7 7	6 5	4 26.8	19.84	6.1404	Synthesis Synthesis	20.6596	0.7708806
Amelung	1998	33.0	5 -97.2	1 19		865	Vertisols	HM	0	10 Gr	assland	Particle_Size	45 3	12 2	3 7	7 59.7	41.76	7.1408	Synthesis	52.5592	0.8803886
Amelung	1998	29.4	2 -96.3	3 3	20	1030	Mollisols	HM	0	10 Gr	assland	Particle_Size	25 1	4 6	1 3	9 23.7	15.12	4.9984	Synthesis	18.7016	0.789097
Amelung	1998	27.4	5 -94.0	4 22	2	700	Mollisols	HM	0	10 Gr	assiand	Particle_Size	26	7 6	4 4 7 3	3 16	12.56	6.6429	Synthesis	9.3571	0.5848188
Amelung	1998	27.5	7 -98.54	4 23	.4	440	Aridisols	HM	0	10 Gr	assland	Particle_Size	28 2	2 5	0 5	0 22.5	22.22	2.6826	Synthesis	19.8174	0.8807733
Anderson	1981	50	-103	5	3	450	Mollisols Mollisols	HM	0	10 Cro 10 Cro	opland	Particle_Size Particle_Size	37 3	14 2 15 3	9 7	1 17.1	16.04	0.64	Synthesis Synthesis	16.46	0.9625731
Angers	1993	NA	NA	NA	NA				0 NA	Cro	opland	Particle_Size	23 5	i0 2	7 7	3 27	17.07	9.8275862	Synthesis	17.172414	0.6360153
Angers	1993	NA	NA	NA	NA				0 NA	Cro	opland	Particle_Size	23 5	50 2 50 2	7 7	3 27.5	17.12	9.9099099	Synthesis	17.59009	0.6396396
Angers	1991	NA	NA	NA	NA		Spodosols	HM	0	15 Cro	opland	Particle_Size	12 5	3 3	5 6	5 27.1	23.27	6.545	Synthesis	20.555	0.7584871
Angers	1991	NA	NA	NA	NA		Spodosols	HM	0	15 Cro	opland	Particle_Size	13 5	6 3	1 6	9 35.2	29.87	8.308	Synthesis	26.892	0.7639773
Balabane	2004	4	9 3	2 :	LO	640	Alfisols	HM	0	15 Cro 15 Fal	llow	Particle_Size	12 1	13 6	1 3	9 4.7	3.91	7.689 NA	Synthesis	#VALUE!	IVALUE!
Balabane	2004	4	9	2	10	640	Alfisols	HM	0	15 Cro	opland	Particle_Size	17 2	2 6	1 3	9 11	8.73	NA	Synthesis	#VALUE!	#VALUE!
Balabane	2004	4	9 3	2 :	10	640	Alfisols	HM	0	15 Fal	opland	Particle_Size Particle_Size	16 5	18 2 10 3	3 6	4 4.7	9.26	NA	Synthesis	#VALUE!	#VALUE!
Balesdent	1998	4	3 0.5	5	13	1200	Inceptisols	нм	0	30 Ter	mperate Fc	Particle_Size	21 4	10 3	9 6	1 52.6	47.4	1.90248	Synthesis	50.69752	0.9638312
Balesdent	1998	43	3 0.5	5	13	1200	Inceptisols	HM	0	26 Cro	opland	Particle_Size	19 4	13 3	8 6	2 30.9	25.8	1.6841	Synthesis	29.2159	0.9454984
Balesdent	1998	4	3 0.	5 3	13	1200	Inceptisols	HM	0	30 Ter	mperate Fc	Particle_Size	21 6	57 1	2 8	8 52.6	48.18	1.11648	Synthesis	51.48352	0.9787741
Balesdent	1998	4	3 0.1	5 :	13	1200	Inceptisols	HM	0	26 Cro	opland	Particle_Size	19 7	72	9 9	1 30.9	26.26	1.22122	Synthesis	29.67878	0.9604783
Balesdent	1998	NA 4	NA U.	NA .	NA	1200	inceptisois	HM	0 NA	26 Cro	assland	Particle_Size	18 19 6	50 2	1 7	8 17.8 9 58.8	30.7	0.92136	Synthesis	10.8/864	0.9482382
Balesdent	1998	NA	NA	NA	NA				0 NA	Gra	assland	Particle_Size	17 5	4 2	9 7	1 31.9	79.6	21.4	Synthesis	10.5	0.3291536
Balesdent	1998	NA	NA	NA	NA NA				0 NA 0 NA	Cro	opland assland	Particle_Size Particle Size	24 3 19 7	19 3	8 9	3 16.9	8.9	1.2	Synthesis Synthesis	15.7	0.9289941
Balesdent	1998	NA	NA	NA	NA				0 NA	Gra	assland	Particle_Size	17 7	18	5 9	5 31.9	98.6	2.4	Synthesis	29.5	0.9247649
Balesdent	1998	NA	NA 12	NA 3	NA 25	1100	Ovisels	IM	0 NA	10 Cro	opland	Particle_Size	24 6	54 1	2 8	8 16.9	9.3	0.8	Synthesis Synthesis	16.1	0.9526627
Barthes	2008		4 13.	3 3	25	1100	Oxisols	LM	0	10 Sav	opland	Particle_Size NA	NA	2	2 7	8 21.8	16.7	3.6	Synthesis	18.2	0.8348624
Barthes	2008	-	4 13.	3	25	1100	Oxisols	LM	0	10 Sav	vanna	Particle_Size NA	NA	2	4 7	6 36.4	27.8	7.7	Synthesis	28.7	0.7884615
Barthes	2008	-4.	1 13.	3 3	25	1400	Oxisols Oxisols	LM	0	10 Sav 10 Cre	opland	Particle_Size_NA Particle_Size_NA	NA	3	3 7	3 42.5	22.4	17	Synthesis	25.5	0.6
Barthes	2008	6.2	4 2.3	2 2	27	1200	Ultisols	LM	0	10 Cro	opland	Particle_Size NA	NA	7	4 2	6 5.1	2.9	1.6	Synthesis	3.5	0.6862745
Barthes	2008	6.2	4 2.	2 2	27	1200	Ultisols	LM	0	10 Cro	opland	Particle_Size_NA	NA	8	6 1	4 6.5	3.3	2.3	Synthesis	4.2	0.6461538
Barthes	2008	6.2	4 2.	2 2	27	1200	Ultisols	LM	0	10 Cro	opland	Particle_Size NA	NA	8	1 1	9 8.5	4.1	2.5	Synthesis	6	0.7058824
Barthes	2008	-10	5 -49.3	3 3	23	1500	Oxisols	LM	0	10 Sav	vanna	Particle_Size NA	NA	5	3 4	7 22.6	15.9	5.3	Synthesis	17.3	0.7654867
Barthes	2008	-10	5 -49.	3 3	23	1500	Oxisols	LM	0	10 Cre 10 Par	sture	Particle_Size NA Particle_Size NA	NA	5	2 4	8 22	17.1	3.2	Synthesis	18.7	0.8545455
Barthes	2008	-21.2	-48.0	3	23	1600	Oxisols	LM	0	10 Cro	opland	Particle_Size NA	NA	4	0 6	0 20.7	12.6	6.4	Synthesis	14.3	0.6908213
Barthes	2008	-21.3	5 -48.2	2 2	23	1500	Ultisols Entisols	LM	0	10 Cro	opland	Particle_Size_NA Particle_Size_NA	NA	8	1 1	9 7.4	5.2	2.2	Synthesis Synthesis	5.2	0.7027027
Barthes	2008	-23.2	3 -51.1	1 2	21	1600	Oxisols	LM	0	10 Tro	opical Fore	Particle_Size NA	NA	2	3 7	7 30.8	17.3	7.8	Synthesis	23	0.7467532
Barthes	2008	-23.2	3 -51.1	1 2	21	1600	Oxisols	LM	0	10 Cro	opland	Particle_Size NA	NA	1	3 8	7 17.8	12.9	2.7	Synthesis	15.1	0.8483146
Bates	1960	-23.2	4 4	4 26	.7	1230	Ultisols	LM	0	10 Tro	opical Fore	Particle_Size	1	9 9	0 1	0 39.5	7.06	24.4	Synthesis	15.0	0.3822785
Besnard	2001	4	9 4	4	10	620	Inceptisols	HM	0	10 Cro	opland	Particle_Size	38	3 5	9 4	1 15.2	8.58	3.79803	Synthesis	11.40197	0.7501296
Besnard Besnard	2001	4	9 4	4 :	10	620	Inceptisols	HM	0	10 Cro 10 Cro	opland	Particle_Size Particle_Size	30 2 29 1	20 5 L3 5	0 5 8 4	2 52.2	12.16	12.6228	Synthesis Synthesis	24.3764	0.5126332
Besnard	2001	4	9 4	4	10	620	Inceptisols	нм	0	10 Cro	opland	Particle_Size	34 2	20 4	6 5	4 28.5	12.3	14.7706	Synthesis	13.7294	0.4817333
Besnard	2001	4	9 4	4 :	10	620	Inceptisols	HM	0	10 Cro	opland	Particle_Size	38 1	13 4	9 5	1 15.2	9.06	3.31803	Synthesis	11.88197	0.7817086
Bonde	1992	-22./	-4/5	z	1	1250	Oxisols	LM	0	12 Iros	pical Fore P	Particle_Size	53 1	5 5	4 6	b 22.4	20.64	/5/16	Synthesis	14.8284	0.6619821
Bonde Bonde	1992	-22.7	-47.5	2	1	1250 1250	Oxisols Oxisols	LM	0	10 Cro 10 Cro	pland P pland P	Particle_Size Particle_Size	58 1 55 1	3 2	2 6	1 15.2 8 14.1	16.46	2.5538	Synthesis Synthesis	12.6462	0.8319868
Bonde	1992	-22.7	-47.5	2	1	1250	Oxisols	LM	0	6 Trop	pical Fore P	Particle_Size	50 20	8 29.	2 70.	8 46.9	33.5408	10.8194	Synthesis	36.0806	0.7693092
Bonde	1992	-22.7	-47.5	2	1	1250	Oxisols	LM	0	12 Trop	pical Fore P	Particle_Size	53 19.	1 27.	9 72.	1 22.4	21.334	6.8786	Synthesis	15.5214	0.6929196
Bonde	1992	-22.7	-47.5	2	1	1250	Oxisols	LM	0	10 Cro	pland P	Particle_Size	55 21	6 23.	4 76.	5 14.1	16.3143	1.8802	Synthesis	12.2198	0.8666525
Caravaca	1999	38	-1	. 1	7	275	Entisols	HM	0	20 Falk	ow P	Particle_Size	31 4	8 2	1 7	9 18.1	12.96	NA	Synthesis	MVALUE!	MVALUE!
Caravaca	1999	38	-1	. 1	7	275	Entisols	HM	0	20 Falk	ow P	Particle_Size	27 4	1 3	2 6	B 17.7	12.89	NA	Synthesis	#VALUE!	#VALUE!
Caravaca	1999	38	-1	. 1	7	275	Entisols	нм	0	20 Falk	ow P	Particle_Size	15 3	0 5	5 4	5 15.9	9.92	NA	Synthesis	#VALUE!	MVALUE!
Caravaca	1999	38	-1	. 1	7	275	Andisols Entisols	HM	0	∠U Falk 20 Falk	ow P ow P	Particle_Size	19 2 26 4	9 5. 0 3.	4 6	5 6.6 5 12.4	4.16	NA	synthesis Synthesis	IVALUE!	#VALUE!
Caravaca	1999	38	-1	. 1	7	275	Aridisols	нм	0	20 Falk	ow P	Particle_Size	16 2	0 6	4 3	6 7.3	3.97	NA	Synthesis	#VALUE!	#VALUE!
Caravaca Caravaca	1999	38	-1	1	7	275	Inceptisols Fatisole	HM	0	20 Falls	ow P	Particle_Size	14 1	6 7i	D 31	7 0.4	5.76	NA	Synthesis Synthesis	#VALUE!	#VALUE!
Caravaca	1999	38	-1	. 1	7	275	Entisols	нм	0	20 Falk	ow P	Particle_Size	21 4	6 3	3 6	7 4.4	3.64	NA	Synthesis	#VALUE!	#VALUE!
Caravaca	1999	38	-1	. 1	7	275	Entisols	HM	0	20 Falk	ow P	Particle_Size	33 4	5 2	2 71	8 6.2	6.02	NA	Synthesis Synthesis	#VALUE!	#VALUE!
Caravaca	1999	38	-1	. 1	, 7	275	Alfisols	нм	0	20 Falk	ow P	Particle_Size	24 2	4 5	2 41	B 7.4	6.43	NA	Synthesis	#VALUE!	#VALUE!
Caravaca	1999	38	-1	. 1	7	275	Entisols	HM	0	20 Falk	ow P	Particle_Size	30 3	0 4	0 6	5.6	3.89	NA	Synthesis	#VALUE!	#VALUE!
Caravaca	1999	38	-1	1	7	275	Mollisols	HM	0	20 Tem	nperate Fc P	Particle_Size	10 3	1 5	9 4	41.1	22.89	NA	Synthesis	#VALUE!	#VALUE!
Caravaca	1999	38	-1	. 1	7	275	Mollisols	нм	0	20 Terr	nperate Fc P	Particle_Size	12 3	2 5	5 4	4 61.1	43.74	NA	Synthesis	#VALUE!	#VALUE!
Caravaca Caravaca	1999	38	-1	1	7	275	Mollisols	HM	0	20 Terr 20 Terr	nperate Fc P nperate Fc P	Particle_Size Particle_Size	12 3 16 7	8 5	D 50	0 60.2 7 45 7	43.88	NA NA	Synthesis Synthesis	#VALUE!	#VALUE!
Caravaca	1999	38	-1	. 1	7	275	Mollisols	нм	0	20 Terr	nperate Fc P	Particle_Size	22 2	3 5	5 4	5 50.6	38.28	NA	Synthesis	#VALUE!	#VALUE!
Carter	2003	45.3	-73.6		6	920	Incept-Entisc	HM	0	10 Falk	ow P	Particle_Size	71 1	9 1	9	30.2	22	8.2	Synthesis Synthesis	22	0.7284768
Carter	2003	47.2	-70	0.	9	866	Incept-Entisc	HM	0	10 Falk	ow P	Particle_Size	47 3	9 1	4 8	5 22.2	17.2	2.8	Synthesis	17.2	0.7747748
Carter	2003	45.5	-64.1	5.	6	1107	Entisols	нм	0	10 Falk	ow P	Particle_Size	18 6	0 2	2 7	B 27.1	17	10.1	Synthesis	17	0.6273063
Carter	2003	45.6	-67.4	4.	7	4143 819	inceptisols Incept-Entisc	HM	0	10 Falk	ow P ow P	Particle_Size	19 5 36 3	5 21	9 7	2 31.9 1 20.3	19.8 17.8	12.1	Synthesis Synthesis	19.8 17.8	0.8768473
Carter	2003	42.1	-82.4	8.	7	819	Incept-Entisc	нм	0	10 Falls	ow P	Particle_Size	36 3	5 2	9 7	1 21.5	18.8	2.7	Synthesis	18.8	0.8744186
Carter	2003	46.4	-71.1	4.	6 9	1127 846	Incept-Entisc	HM	0	10 Falk	ow P	Particle_Size	28 4	1 3	1 6	9 22.3 8 27 2	18.4	3.9	Synthesis Synthesis	18.4	0.8251121
Carter	2003	45.3	-73.6		6	920	Alfisols	нм	0	10 Falk	ow P	Particle_Size	28 2	2 5	0 51	0 14.5	10.2	4.9	Synthesis	10.2	0.7034483
Carter	2003	46.2	-63.1	5.	9	1077	Spodosols	HM	0	10 Falk	ow P	Particle_Size	8 3	0 6	2 3	8 21.2	18.4	2.8	Synthesis	18.4	0.8679245
Carter	2003	46.2	-63.1	5.	9	846	Inceptisols	HM	0	10 Falk	ow P	Particle_Size	0 3 10 2	5 6	5 3	5 1/.8	15.7	2.1	Synthesis	15.7	0.8820225
Carter	2003	42.5	-80.3	7.	8	935	Alfisols	HM	0	10 Falk	ow P	Particle_Size	6	6 8	8 1	2 7.6	6.2	1.4	Synthesis	6.2	0.8157895
Chan	1987 2001	niA -31.5	niA 147	NA 1	NA B	431	Alfisols	HM	0 NA	10 Pas	ture P	Particle_Size	15 5 32 1	3 3	5 71	5 32.9 5 21.4	26.7	5.3934 NA	synthesis Synthesis	27.5066 #VALUE!	IVALUE!
Chan	2001	-31.5	147	1	8	431	Alfisols	HM	0	10 Cro	pland P	Particle_Size	32 1	3 5	5 4	5 7.5	4.7	NA	Synthesis	#VALUE!	MVALUE!
Chan	2001	-31.5	147	1	8 6	431 550	Alfisols	HM	0	10 Pas 10 Cro	pland pland	Particle_Size	3Z 1 23 1	3 5	5 3	5 8.8 5 17.5	5.2	NA	synthesis Synthesis	IVALUE!	#VALUE!
Chan	2001	-35	147.5	1	6	550	Alfisols	нм	0	10 Cro	pland P	Particle_Size	23 1	2 6	5 3	5 14.8	8.2	NA	Synthesis	#VALUE!	#VALUE!
Chan Chan	2001	-34	148.7	1	5	564 564	Alfisols	HM	0	10 Pas 10 Cm	ture Poland P	Particle_Size Particle_Size	13 1 13 1	1 70	5 2	4 32.2 4 16.º	8.3	NA	Synthesis Synthesis	#VALUE!	#VALUE!
Chan	2001	-34	148.7	1	5	564	Alfisols	нм	0	10 Cro	pland P	Particle_Size	13 1	1 7	6 2	4 6.9	3.5	NA	Synthesis	#VALUE!	/VALUE!
Chan	2001	-30	148	19.	5	474	Vertisols	HM	0	10 Pas	ture P	Particle_Size	58 1	8 2	4 70	6 9.3 6 10	5.6	NA	Synthesis Synthesis	MVALUE!	MVALUE!
Chan	2001	-30	148	. 19.	7	1300	Oxisols	LM	0	10 Cro	pland P	Particle_Size	16	7 7	7 2	3 21.9	8.9	NA	Synthesis	#VALUE!	/VALUE!
Chan	2001	2.2	464	4		1200	Ordente			10 0	allowed in the	and the street					0.7		Country and a second second	10000000000	AN/ALLIET

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