



# Rapid Migration of Radionuclides Leaked from High-Level Waste Tanks: Effects of Salinity Gradients, Wetted Path Geometry and Water Vapor Transport



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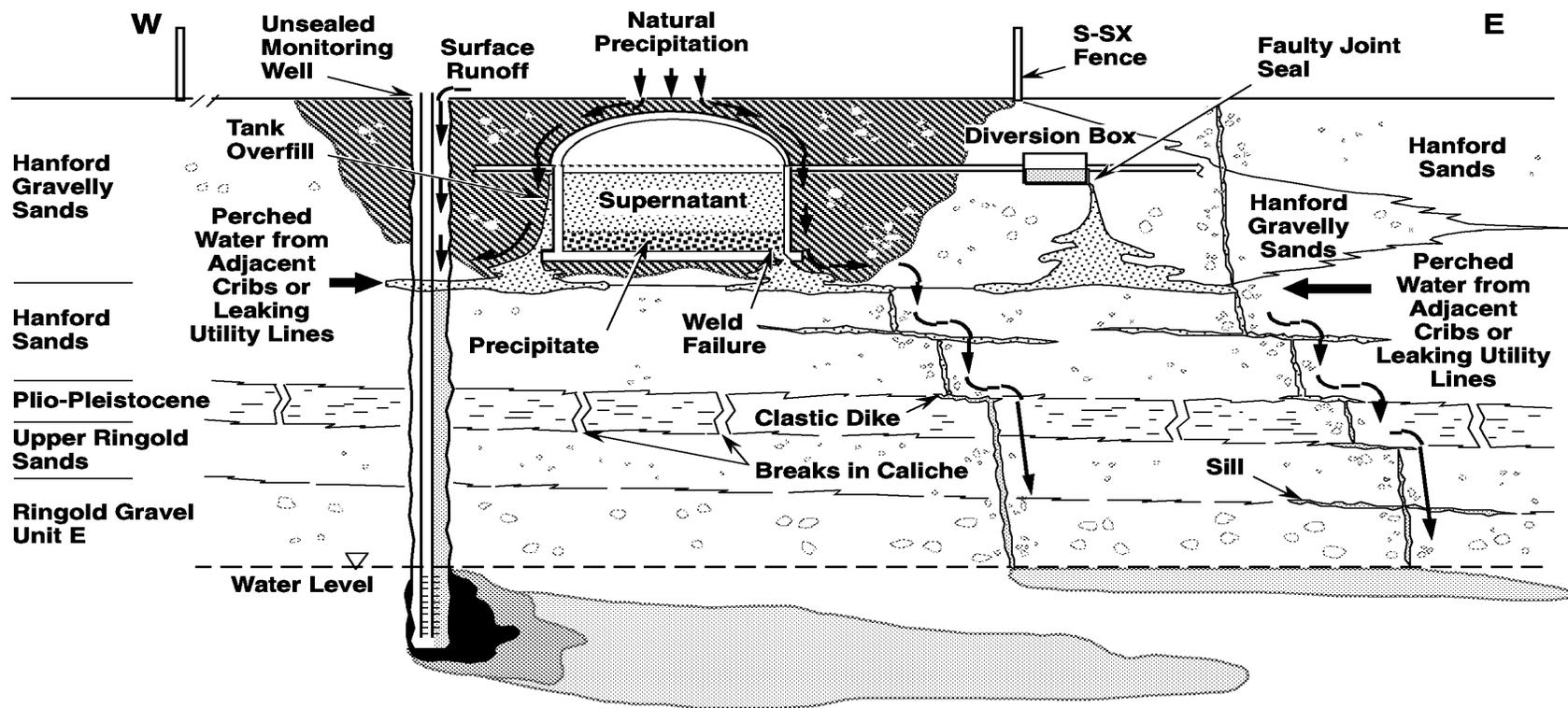
# Background

- 177 High Level Waste Tanks (67 of 146 Single Shell Tanks Known or Suspected to Have Leaked)
- much uncertainty in leak volumes
- monitoring accounts for < 1% of total inventory
- $^{137}\text{Cs}$  "deeper than expected"
- some mobile contaminants in the ground water

# Problem

- Traditional vadose zone transport theory and modeling approaches appear inadequate for explaining observed distribution of contaminants
  - episodic infiltration events
  - fluid properties
  - small-scale variability in lithology
  - enhanced vapor flux
  - unstable wetting fronts

# Hypothetical Sources and Potential Pathways



	$^{137}\text{Cs}$		$^{90}\text{Sr}$		Backfill		Mobile Constituents in Tank Waste ( $^3\text{H}$ , $\text{Na}^+$ , $\text{TcO}_4^-$ , $\text{UO}_2(\text{CO}_3)_3^-$ , $\text{CrO}_4^-$ , $\text{NO}_3^-$ )
	High-Level Waste Supernatant or Liquid Phase				Natural Precipitation and Movement Along Culturally Disturbed and Natural Pathways		
	Solids Phase of High-Level Waste ( $\text{Fe}(\text{OH})_3$ , etc. and Other Precipitated Phases)				Surface Runoff and Artificial Sources of Water		

# Need

- basic research in vadose zone hydrology
  - understand subsurface flow/transport
  - improve effectiveness of remediation
  - calculate risk associated with different clean up strategies

# Scope This Study

To investigate the Effects of Fluid Properties

- 54 million gallons of radioactive and hazardous waste
- caustic:  $10 < \text{pH} < 14$
- saline:  $6 < N < 10$
- dense: specific gravity  $\geq 1.4$

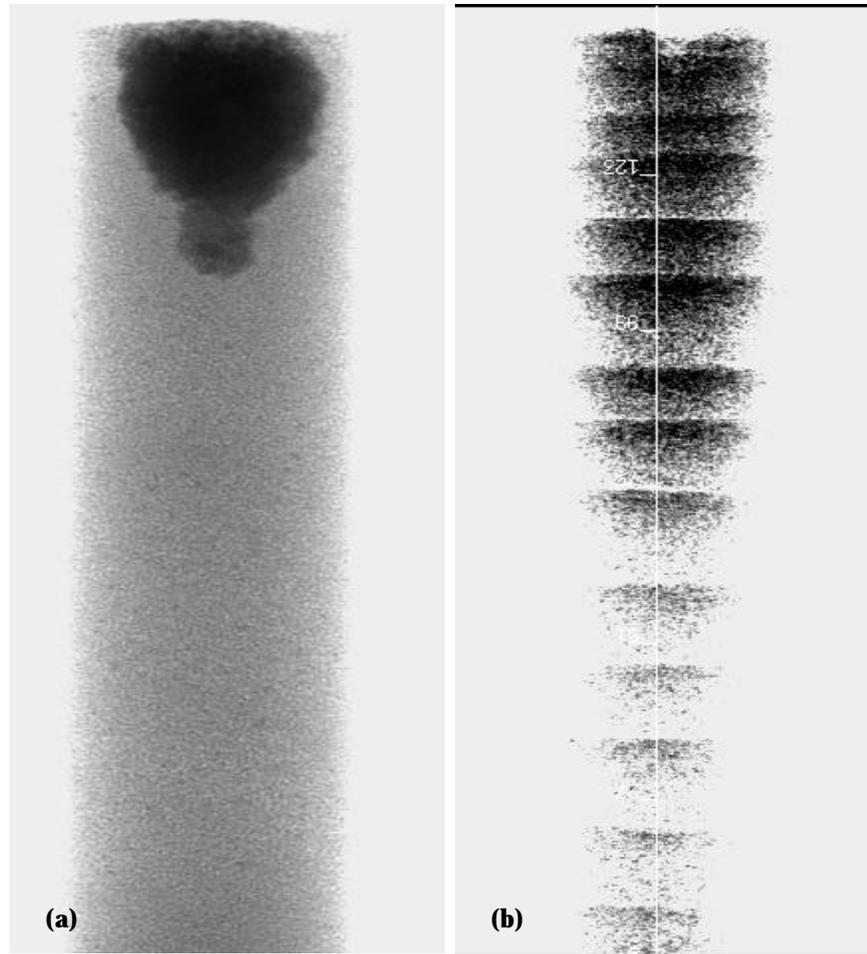
# Objectives

- study the effect of elevated surface tension of caustic brines on flow and transport
- determine the conditions under which osmotically-enhanced vapor flux operate and quantify impact on transport processes
- develop and incorporate theory describing these processes into STOMP for prediction at realistic spatial and temporal scales

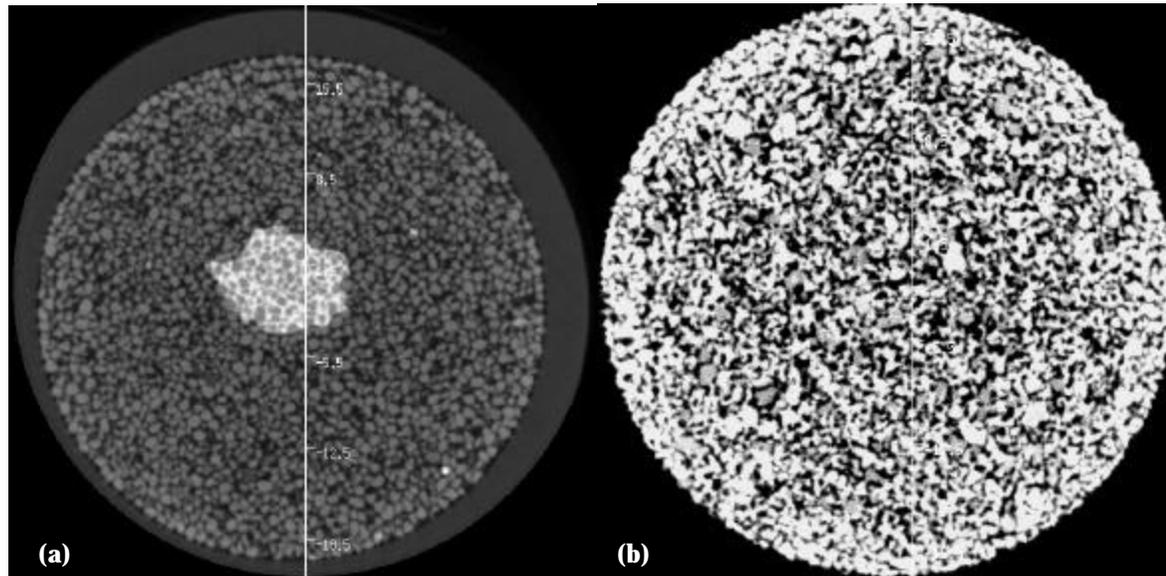
# Collaborators

- Oregon State University (OSU)
  - laboratory scale experiments with Miller Sands
- Desert Research Institute (DRI)
  - effects of hypersaline fluids on hydraulic properties
- Pacific NW National Laboratory (PNNL)
  - laboratory experiments with Hanford Sediments
  - field experiments
  - modeling

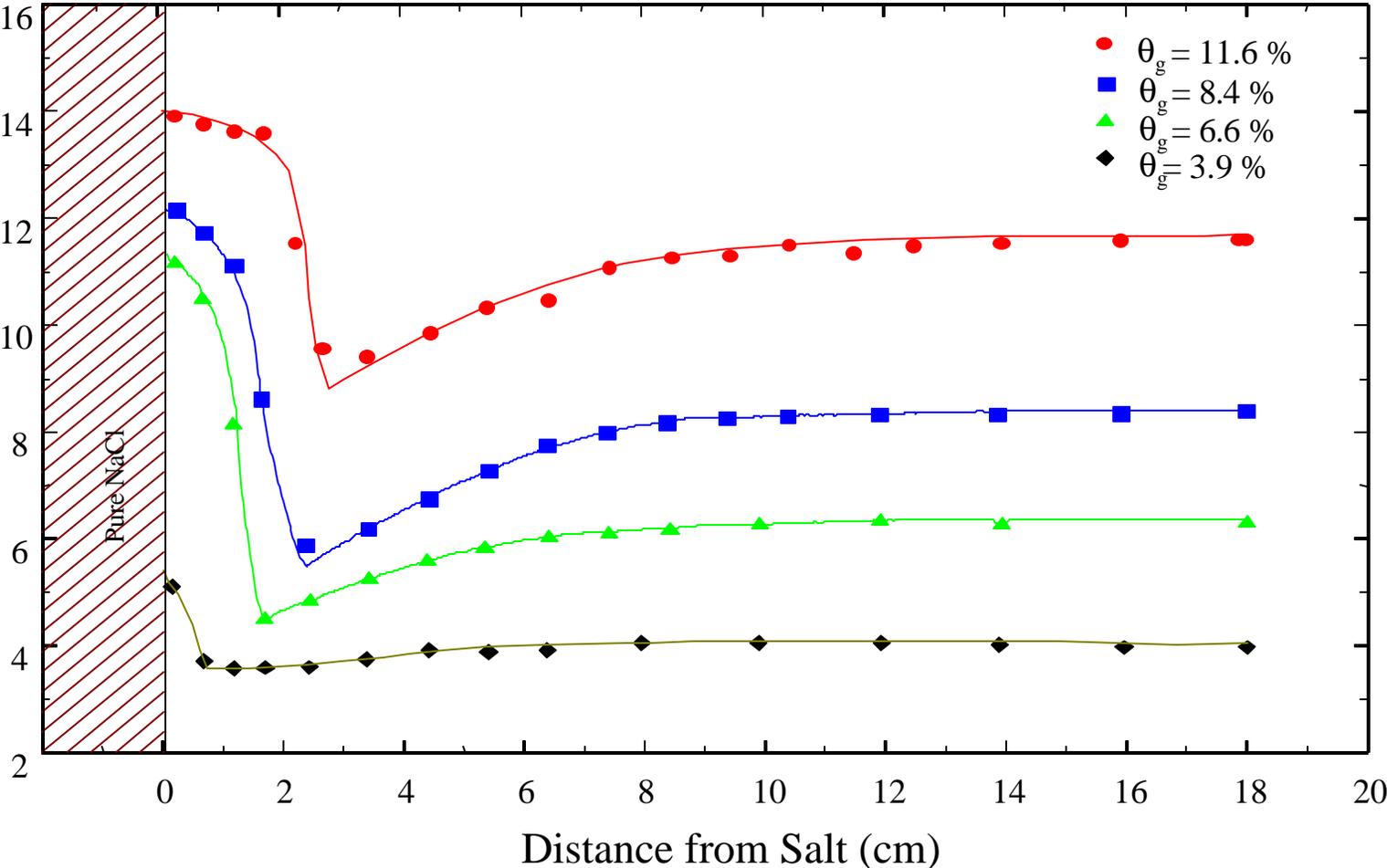
# Xray CT Studies- Unstable Flow



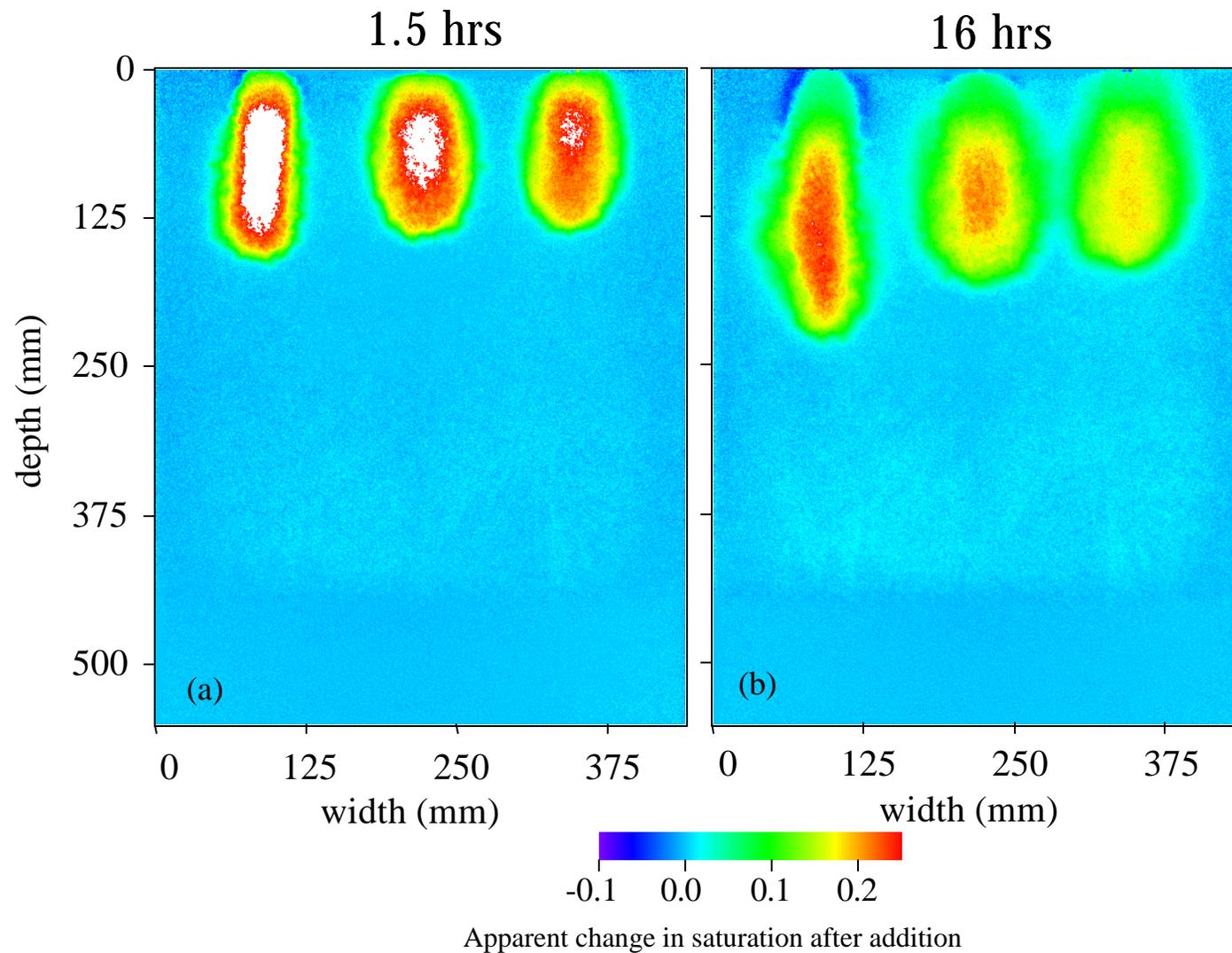
# X-ray CT- Unstable Flow



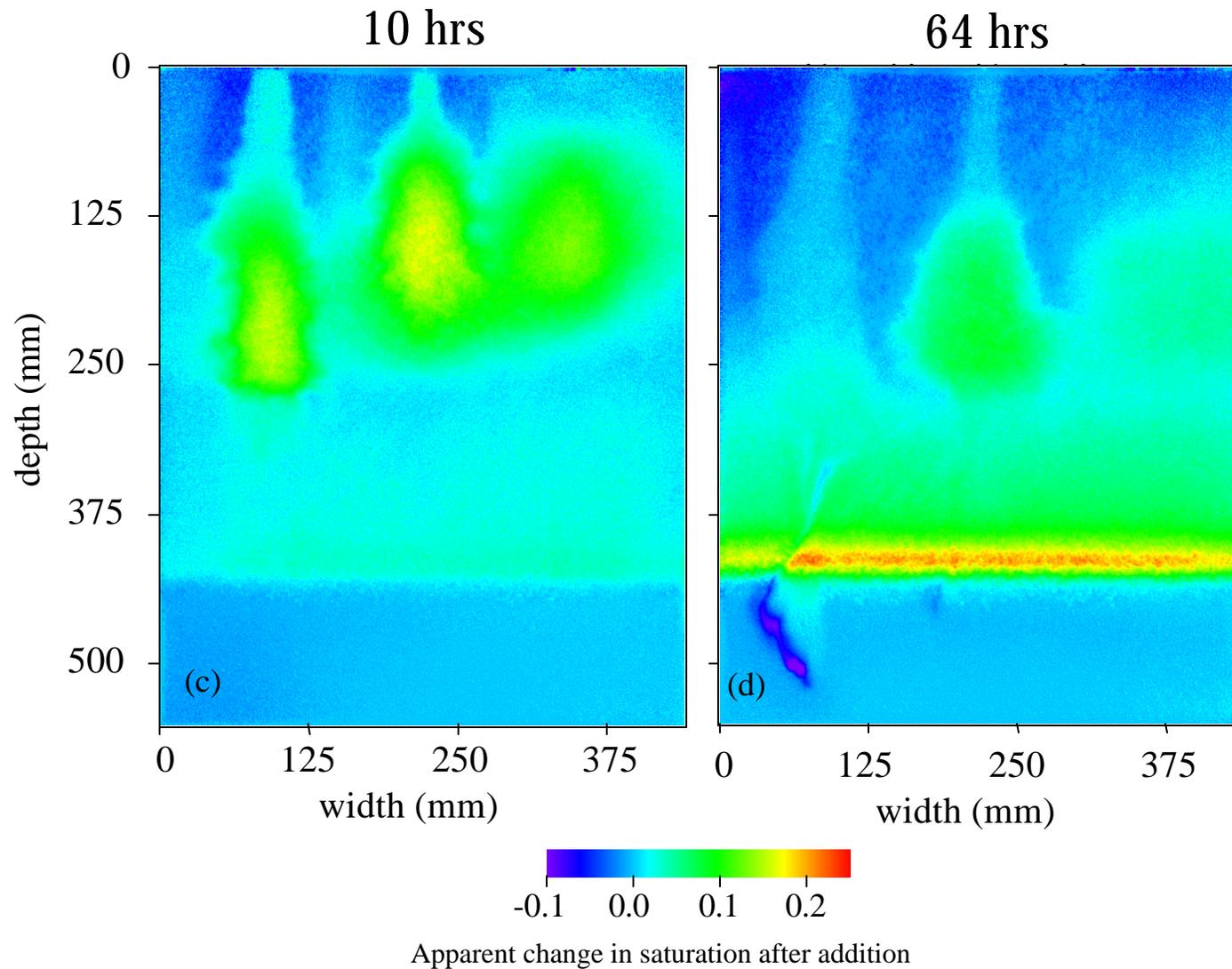
# Osmotic Effects of Moisture Distribution



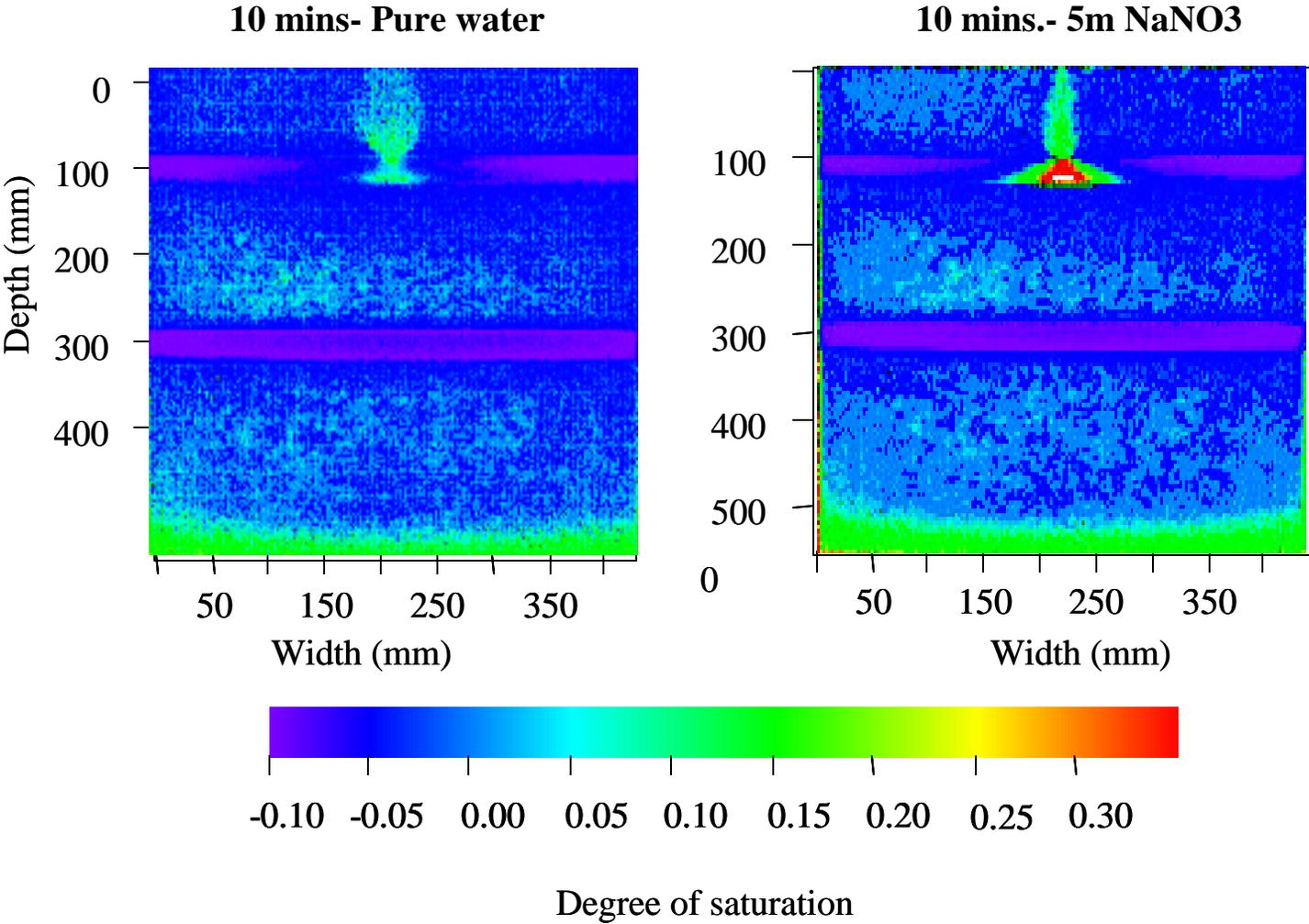
# Laboratory Studies- Homogeneous Sands



# Laboratory Studies- Homogeneous Sands



# Laboratory Studies- Layered Sands



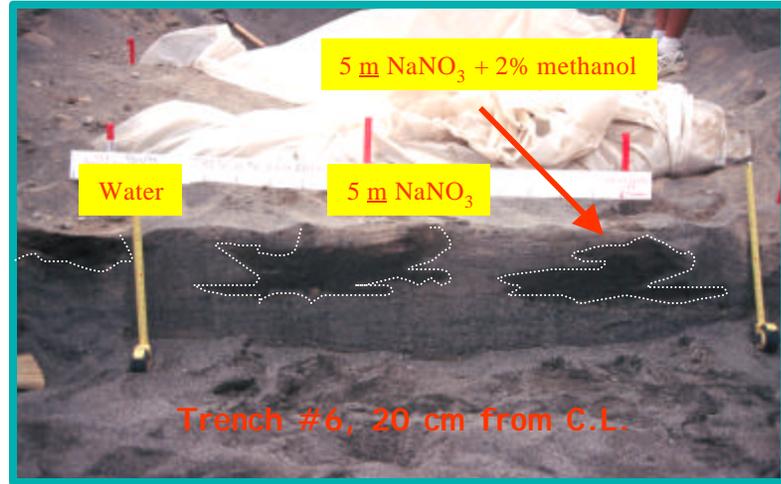
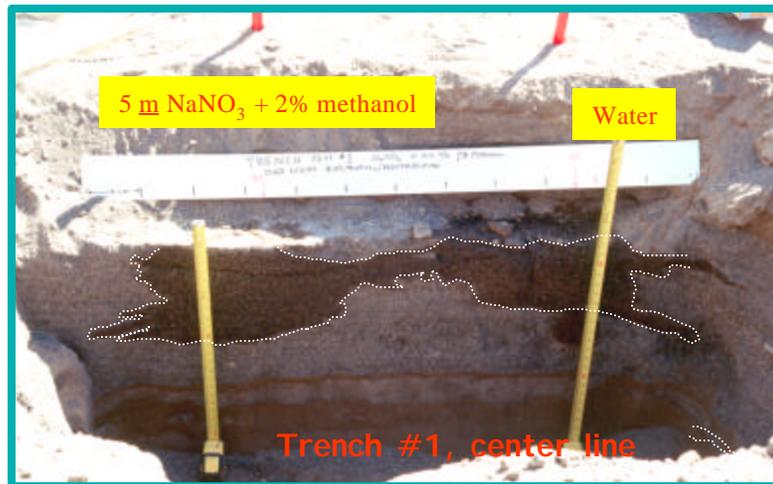
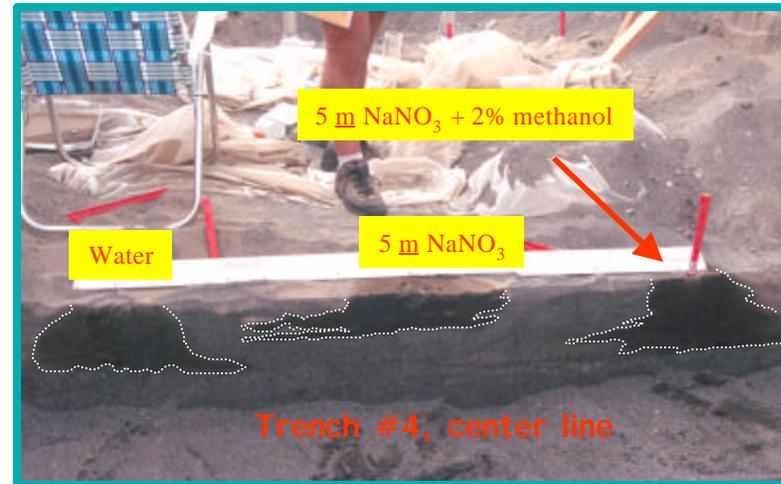
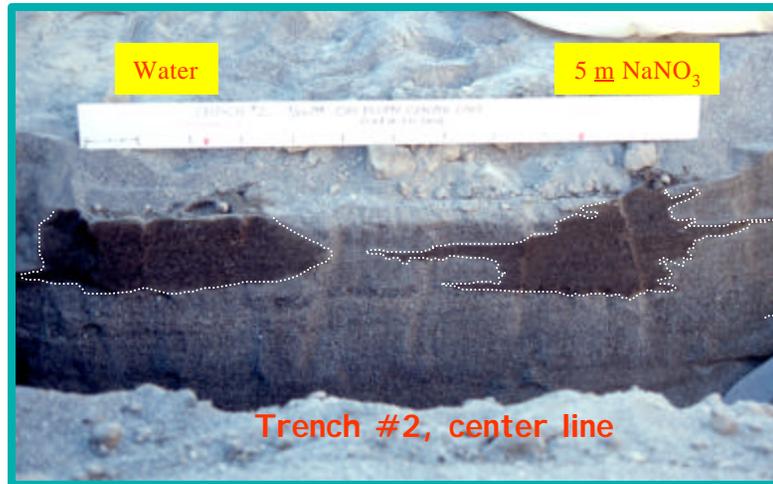
# Field Infiltration Tests

- Six reps
- Solutions applied at 0.4 L/ hr over 4 hrs with a tension infiltrometer
- Trench excavated between 6 and 12 hrs after application

# Typical Profile at Field Site



# Field Studies



# Plume Characteristics

<b><u>Solution</u></b>	<b><u>Mean depth (cm)</u></b>	<b><u>Mean width (cm)</u></b>	<b><u>Aspect ratio</u></b>
<b>5 <u>m</u> NaNO<sub>3</sub></b>	<b>17.5 [3.7]</b>	<b>75.2 [6.5]</b>	<b>4.43 [0.87]</b>
<b>5 m NaNO<sub>3</sub> + 2% methanol</b>	<b>22.75 [2.4]</b>	<b>67.5 [3.5]</b>	<b>2.99 [0.48]</b>
<b>Water</b>	<b>20.25 [5.9]</b>	<b>58.3 [10.3]</b>	<b>3.16 [1.20]</b>

# Thermodynamic Model for Surface Tension of Concentrated Electrolytes

Gibbs thermodynamic, or phenomenological surface-phase, method (Butler, 1932). For salt MX in water (component  $i = M, X, w$ ):

$$m_i^B = m_i^{BO} + RT \ln a_i^B$$

$$m_i^S = m_i^{SO} + RT \ln a_i^S - \bar{A}_i g_{sol}$$

Assumptions:

$$m_w^B = m_w^S \text{ at equilibrium}$$

$$A_w = \bar{A}_w = V_w^{2/3} N^{1/3}$$

$$m^S = g m^B$$

Vapor Phase
Surface Phase MX + H <sub>2</sub> O
Liquid Phase MX + H <sub>2</sub> O

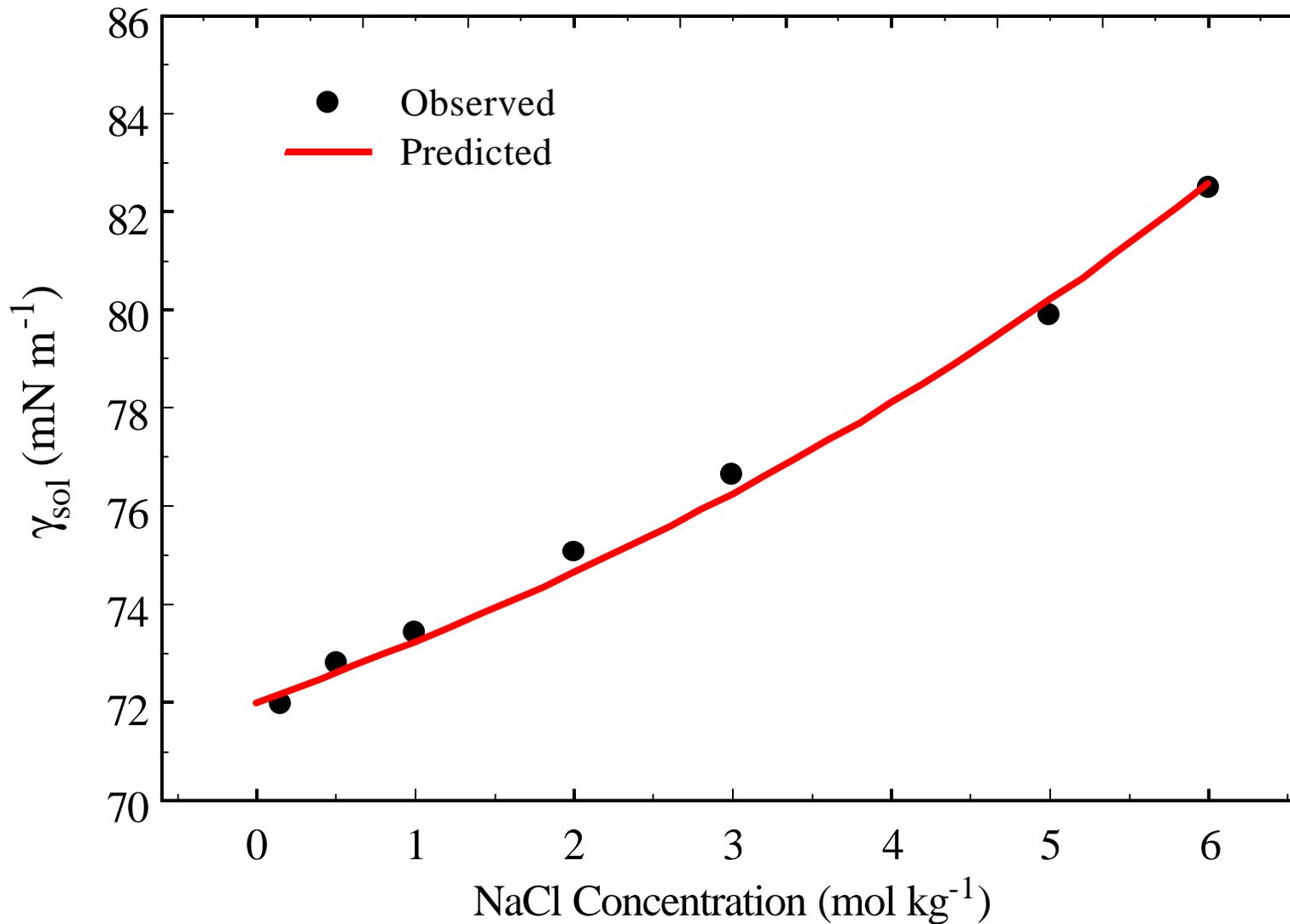
# Electrolyte Surface Tension

$$g_{\text{soln}} = g_w + \frac{RT}{A_w M_w} \left( f^B \sum v_i m_i^B - f^S \sum g_i v_i m_i^B \right)$$

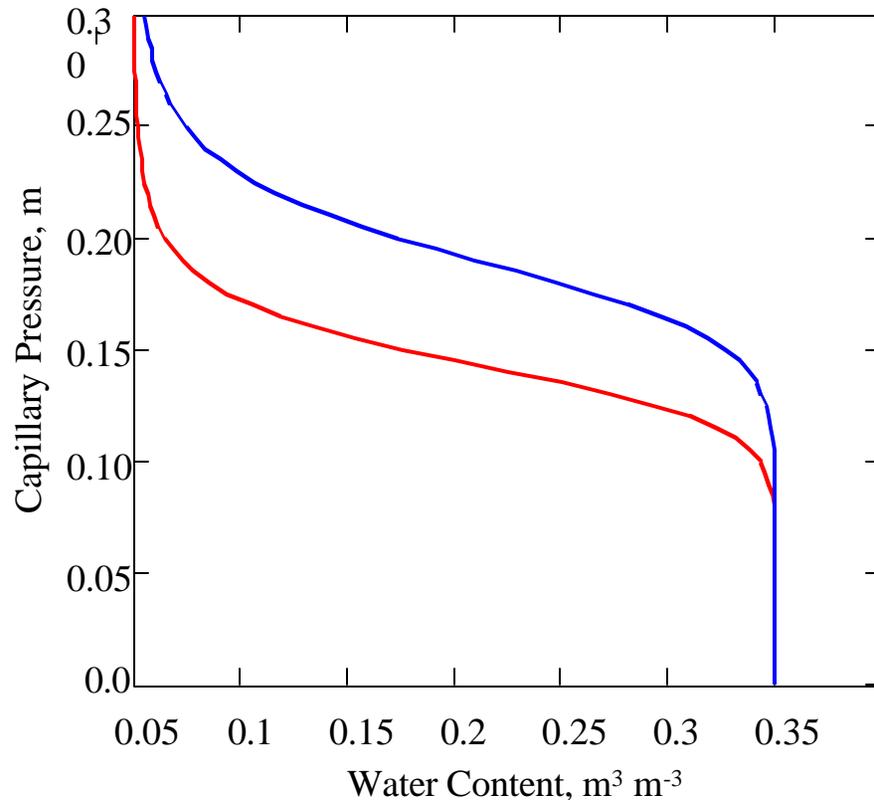
Osmotic Coefficient, Pitzer (1973):

$$f = 1 - |z_M z_X| f^f + m(2v_M v_X / \nu) B_{MX}^f \dots$$
$$+ m^2 [2(v_M v_X)^{3/2} / \nu] C_{MX}^f$$

# Observed vs Predicted $\gamma_{\text{sol}}$ for NaCl

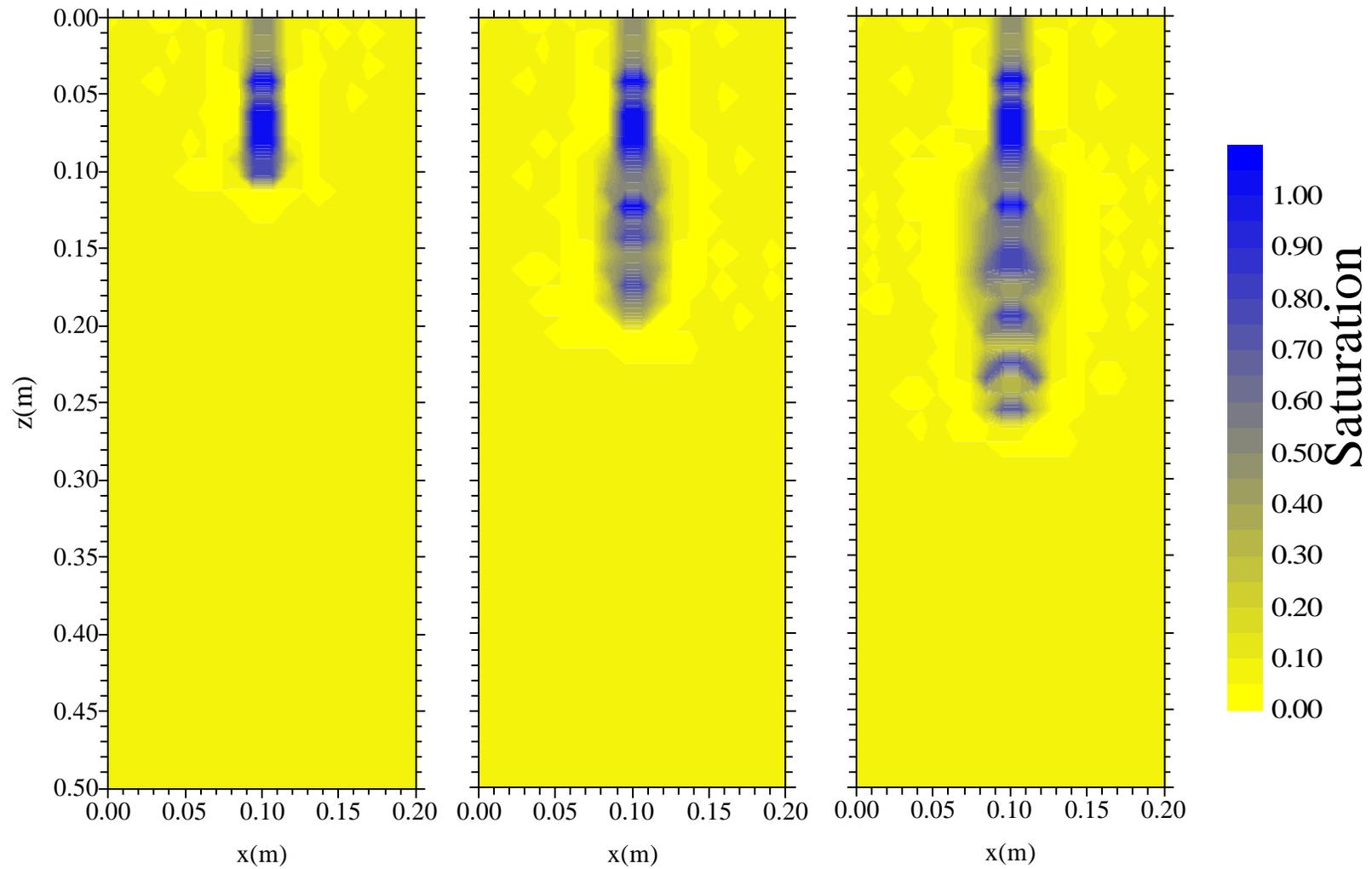


# Effect of $\gamma_{sol}$ on $\psi(\theta)$

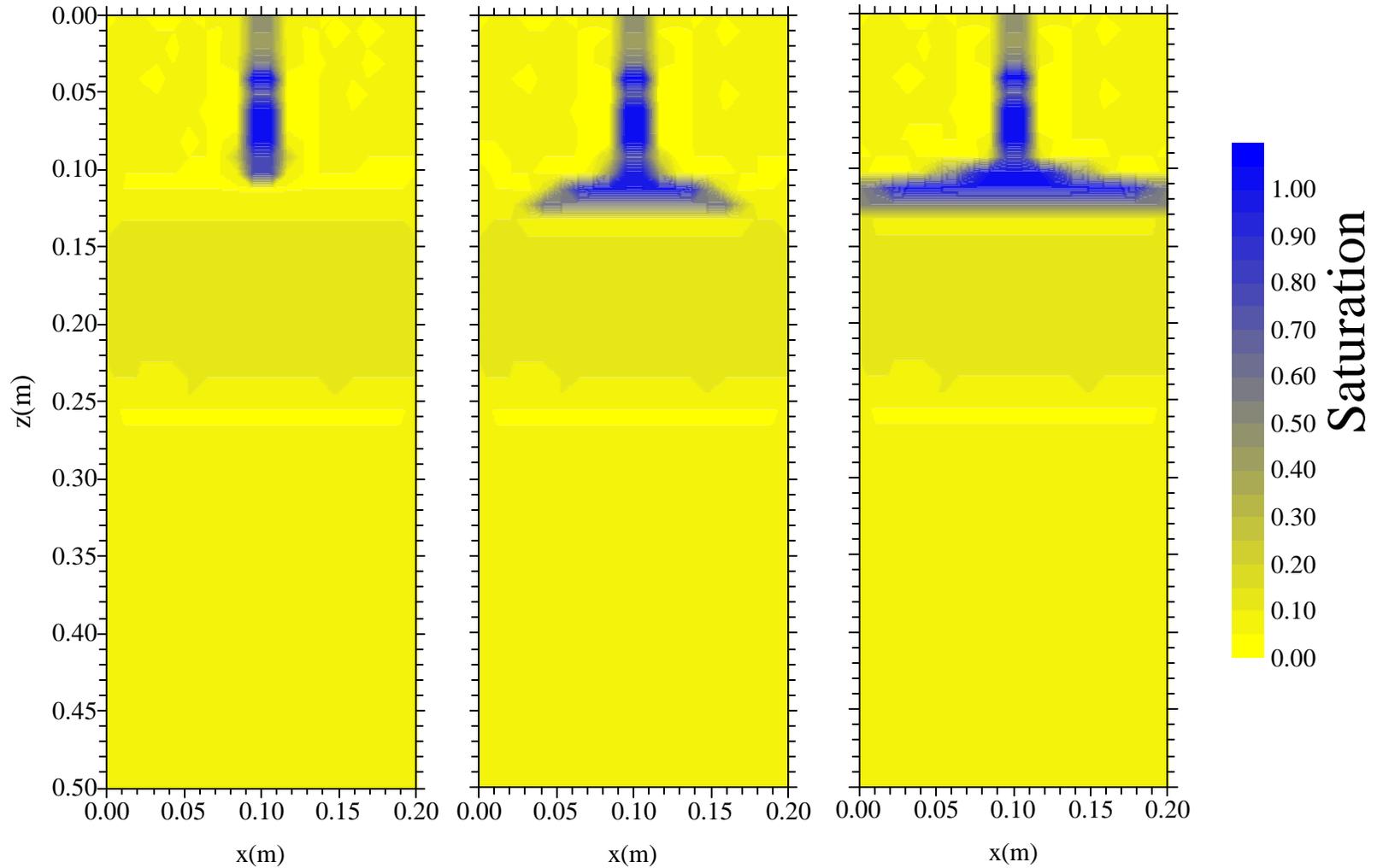


- Linear scaling in  $\psi(\theta)$  cannot reproduce lab observations
- Contact angle approach in which  $\alpha_a > \alpha_r$  producing desired results
  - Differences not constant over entire  $\theta$  but greatest at low  $\theta$
  - Little effect on main drainage branch
  - Effect appear to be greatest on main imbibition branch

# STOMP Simulations-Homogenous Column



# STOMP Simulations- Layered Column



# Vapor Flux Contribution

$$J_v = -Dtq_a \frac{\partial r_v}{\partial T} \nabla T - Dtq_a \frac{\partial r_v}{\partial y_m} \nabla y_m - Dtq_a \frac{\partial r_v}{\partial y_o} \nabla y_o$$

$J_v$  = mass flow rate of vapor

$D$  = molecular diffusion coefficient of water vapor in air

$\tau$  = tortuosity of air-filled pore space

$\rho_v$  = water vapor density

$T$  = Temperature

$\Psi_m$  = matric pressure head

$\Psi_o$  = solution osmotic pressure head

# Status

- Laboratory Studies
  - Homogeneous Miller Sands ✓
  - Heterogeneous Miller Sands- in progress
  - Hanford sediments- in progress
  - Effect of hypersaline fluids on  $\psi(\theta)$ - in progress
- Incorporate effects of electrolytes on surface tension ✓
- Implement adaptive gridding ✓
- Effect of Organics ✓
- Vapor flux due to osmotic potential gradient- in progress

# Summary

- Traditional vadose zone theory inadequate for explaining laboratory and field observations
- research being conducted to determine effects of fluid properties on vadose zone transport
- Laboratory observations suggest impact of surface tension and density; theory incorporated into STOMP
- Product- a tool that DOE can use for prediction and evaluation of different SST retrieval/remediation strategies